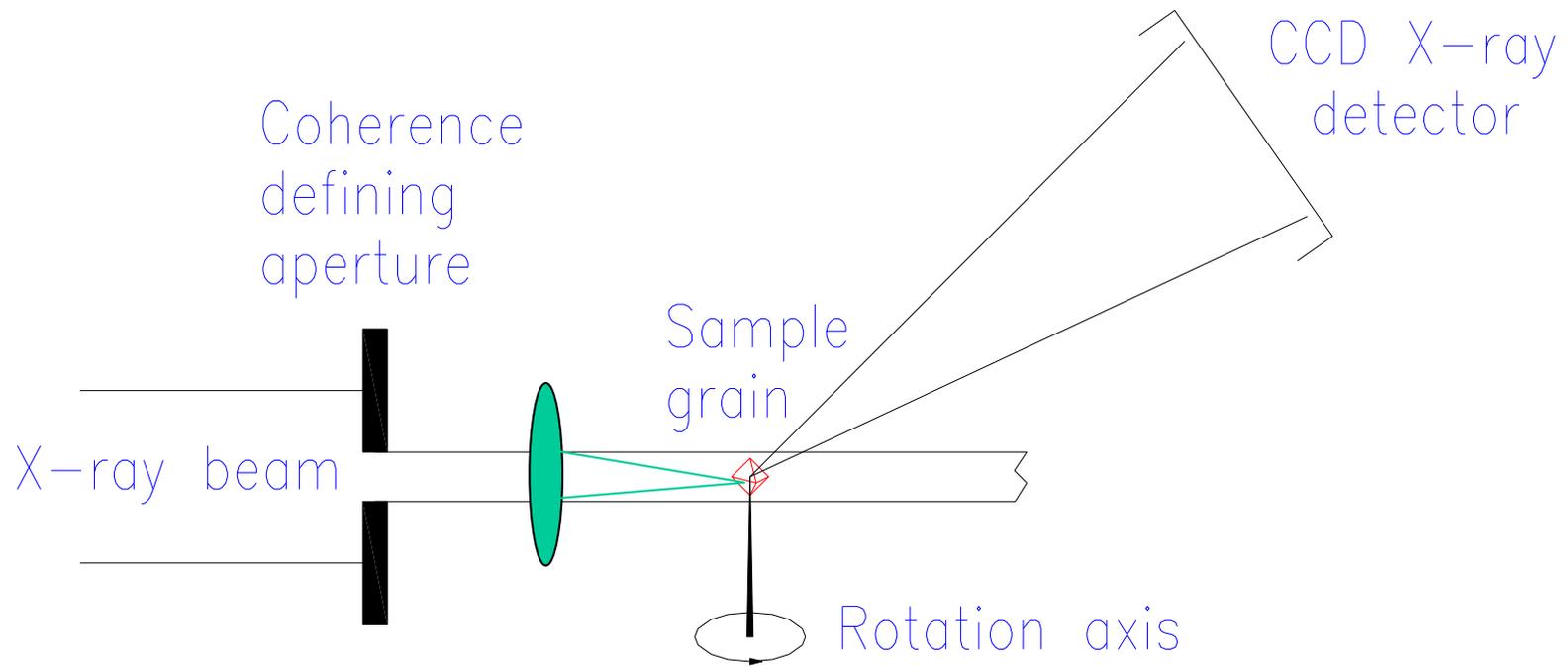


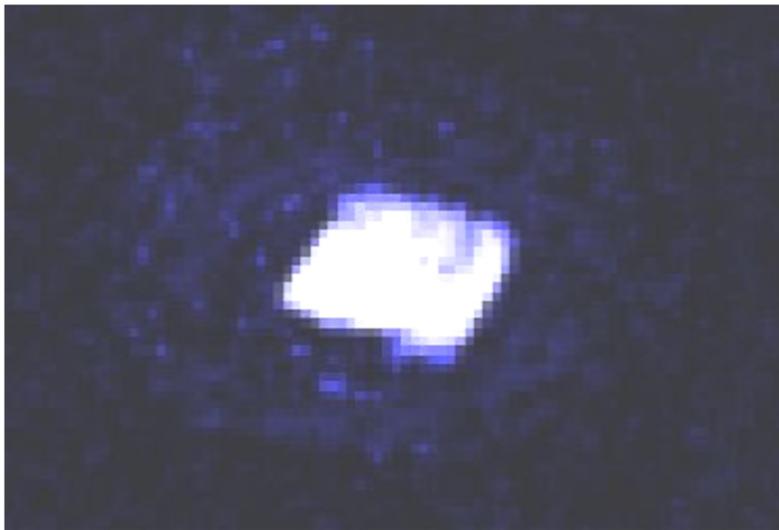
# Solution Scattering Studies of Protein Crystallization

- Ian Robinson
  - Sébastien Boutet
  - Jing Tao
  - Jim Zuo
- Departments of Physics  
and Materials Science  
University of Illinois
- “Workshop on Emerging Areas of  
Biological Crystallography”  
Argonne, July 28 2004

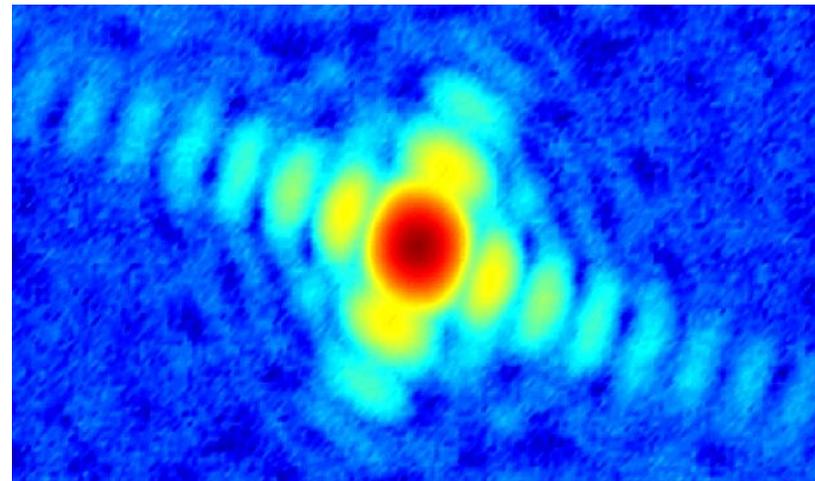
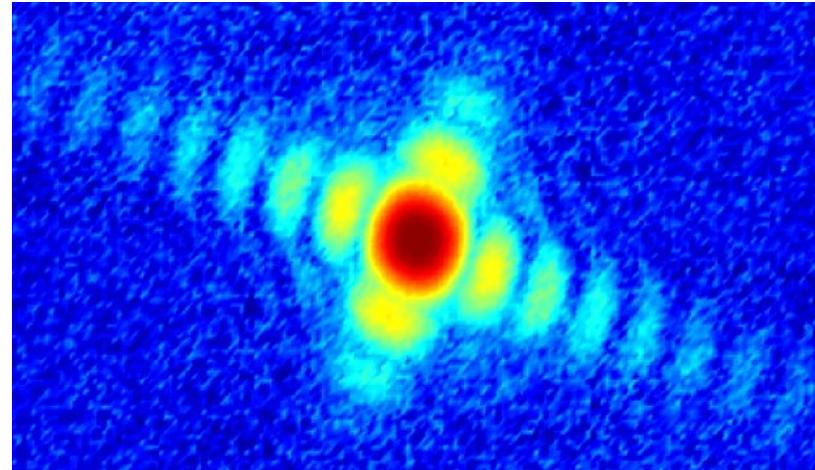
# Lensless X-ray Microscope



# Reconstruction of Ag Nanocrystal

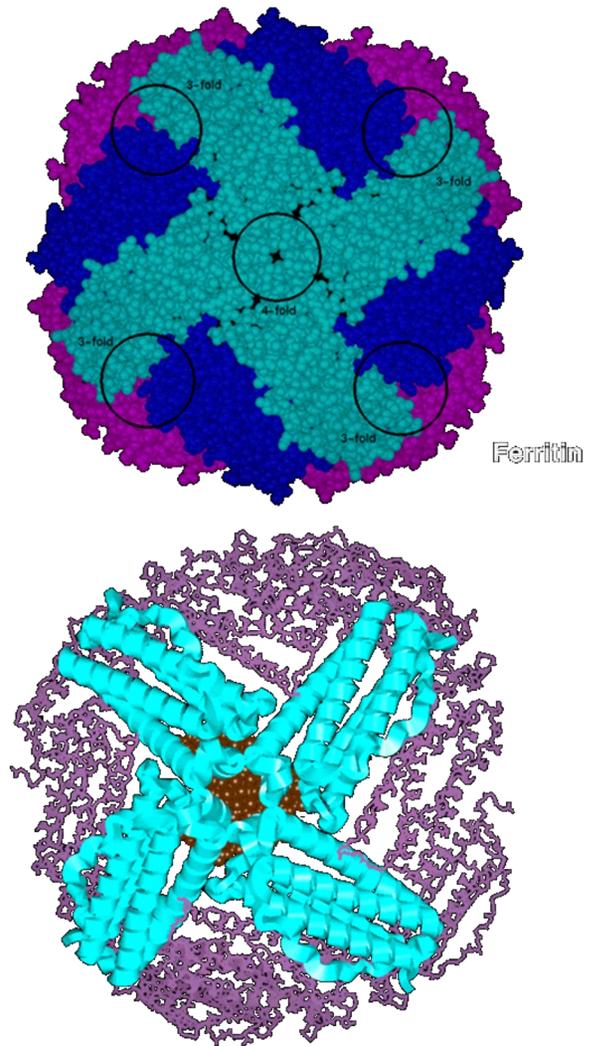


←→  
200nm

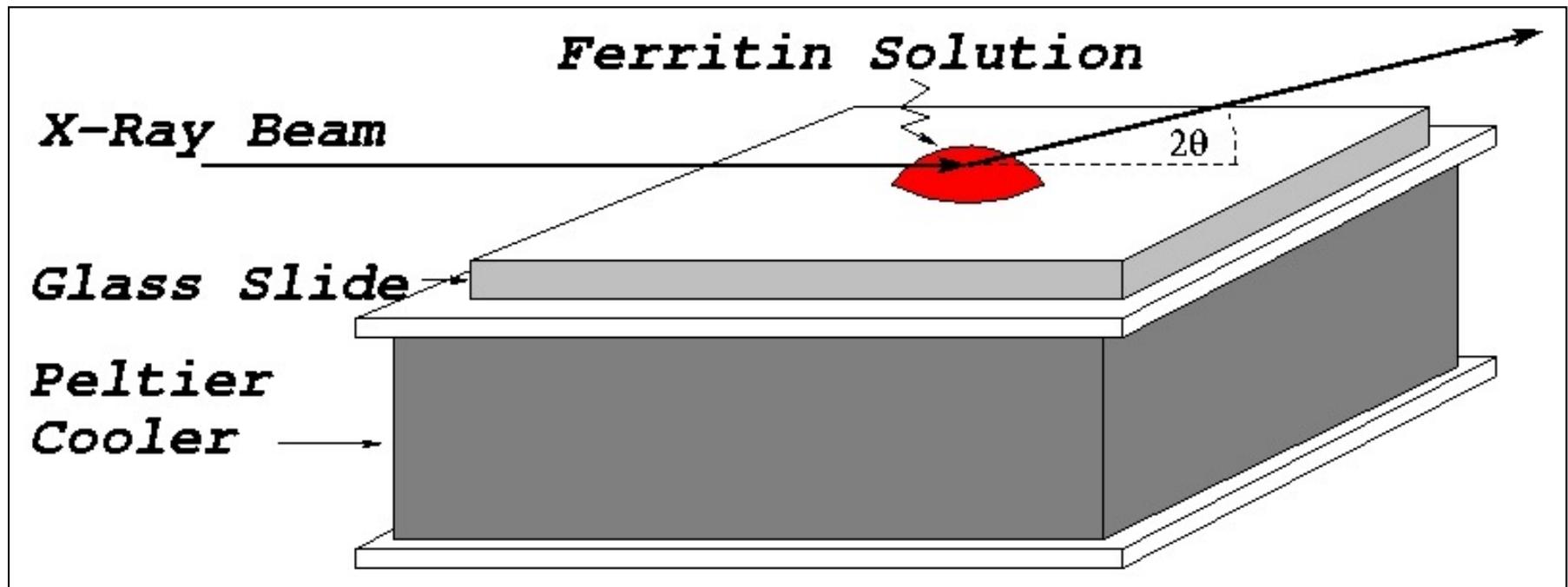


# Horse Spleen Ferritin

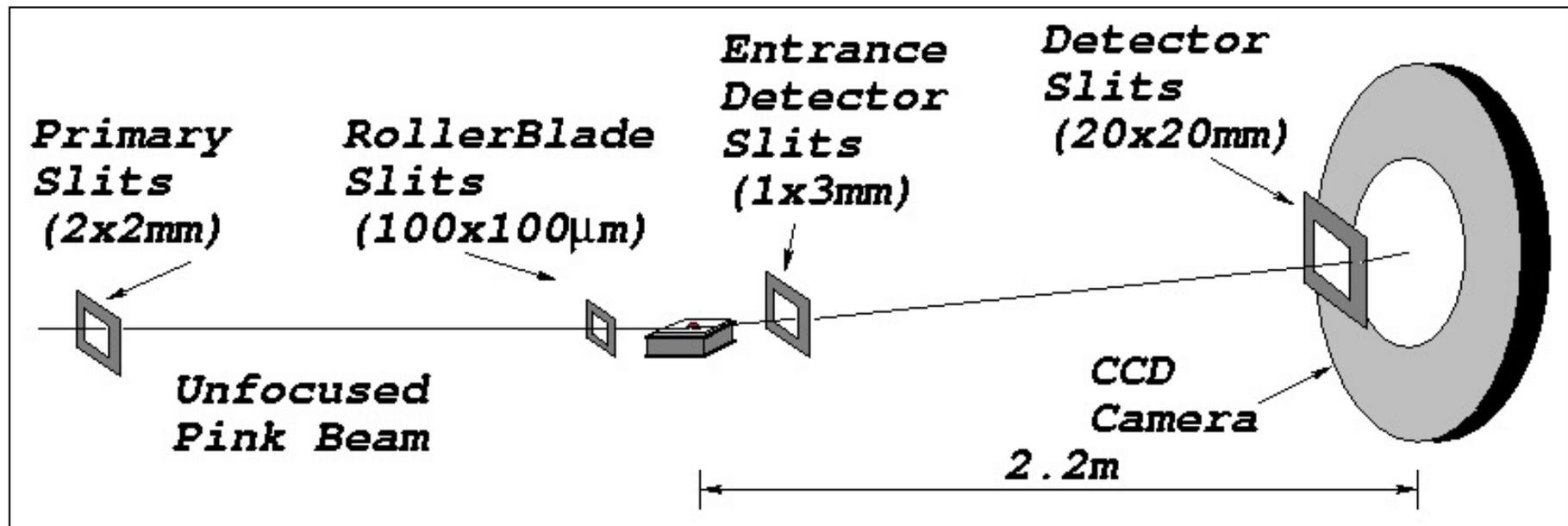
- Ferritin used in almost all living things for Iron Storage
- Made of 24 identical protein subunits arranged in 432 symmetry
- Crystallizes as FCC (I23)
- Inner shell diameter 80Å
- Outer shell diameter 130Å
- Iron core: ferrihydrite form
- With Iron : **Holo**ferritin
- Empty Shell : **Apo**ferritin



# In-situ Study of Crystallization

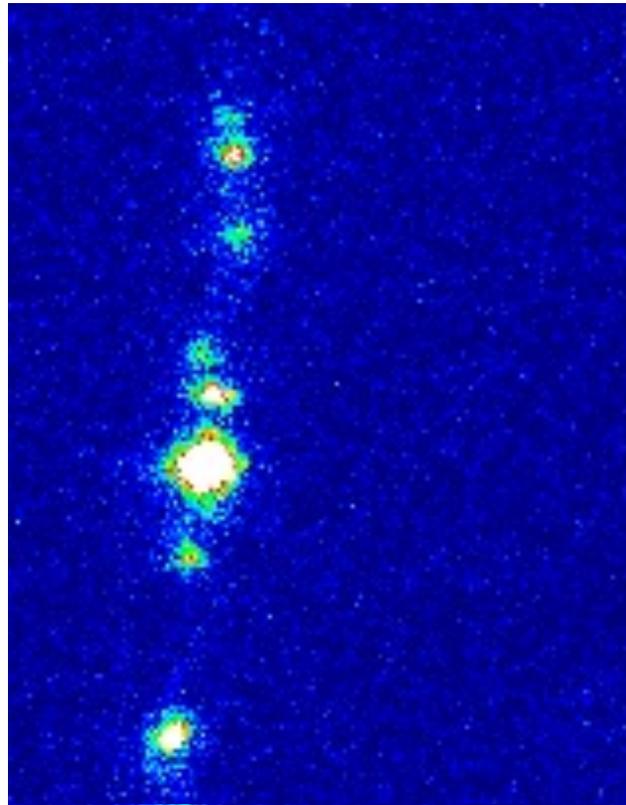


# Experiment at APS Sector 34



# Ferritin (111) Powder Ring

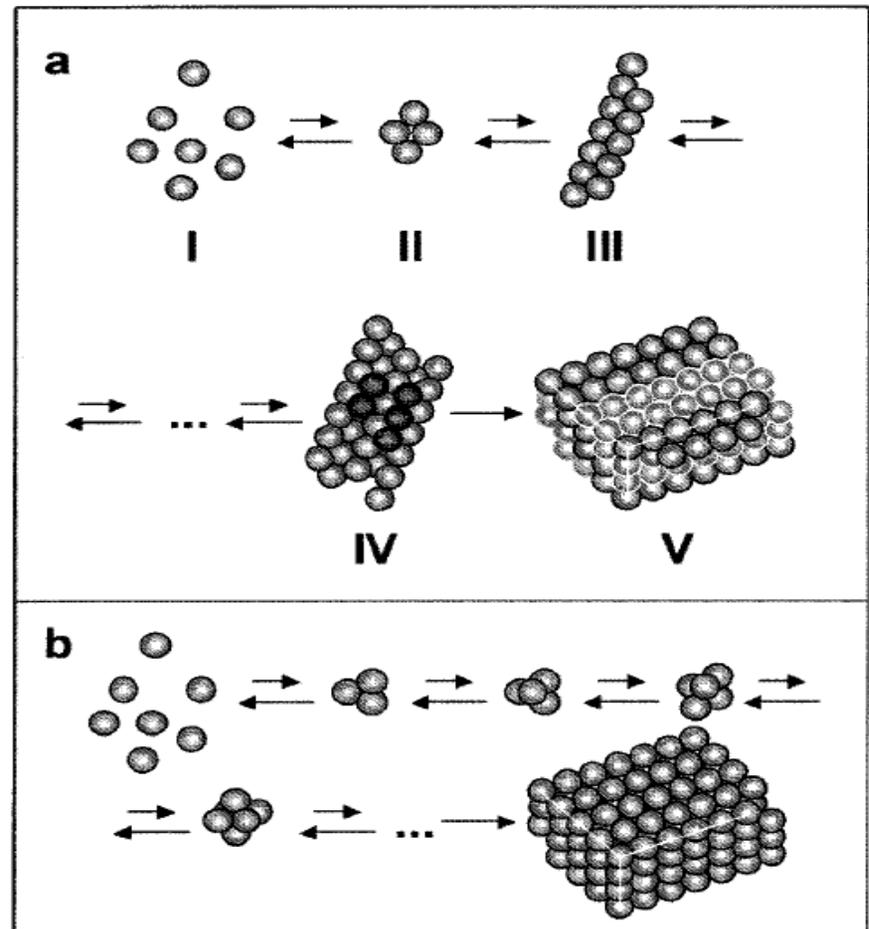
- 50 frames
- 30sec exposure
- 0.3sec playback
- 150x200 pixels  
of 22.5 $\mu\text{m}$



# Critical Nucleation

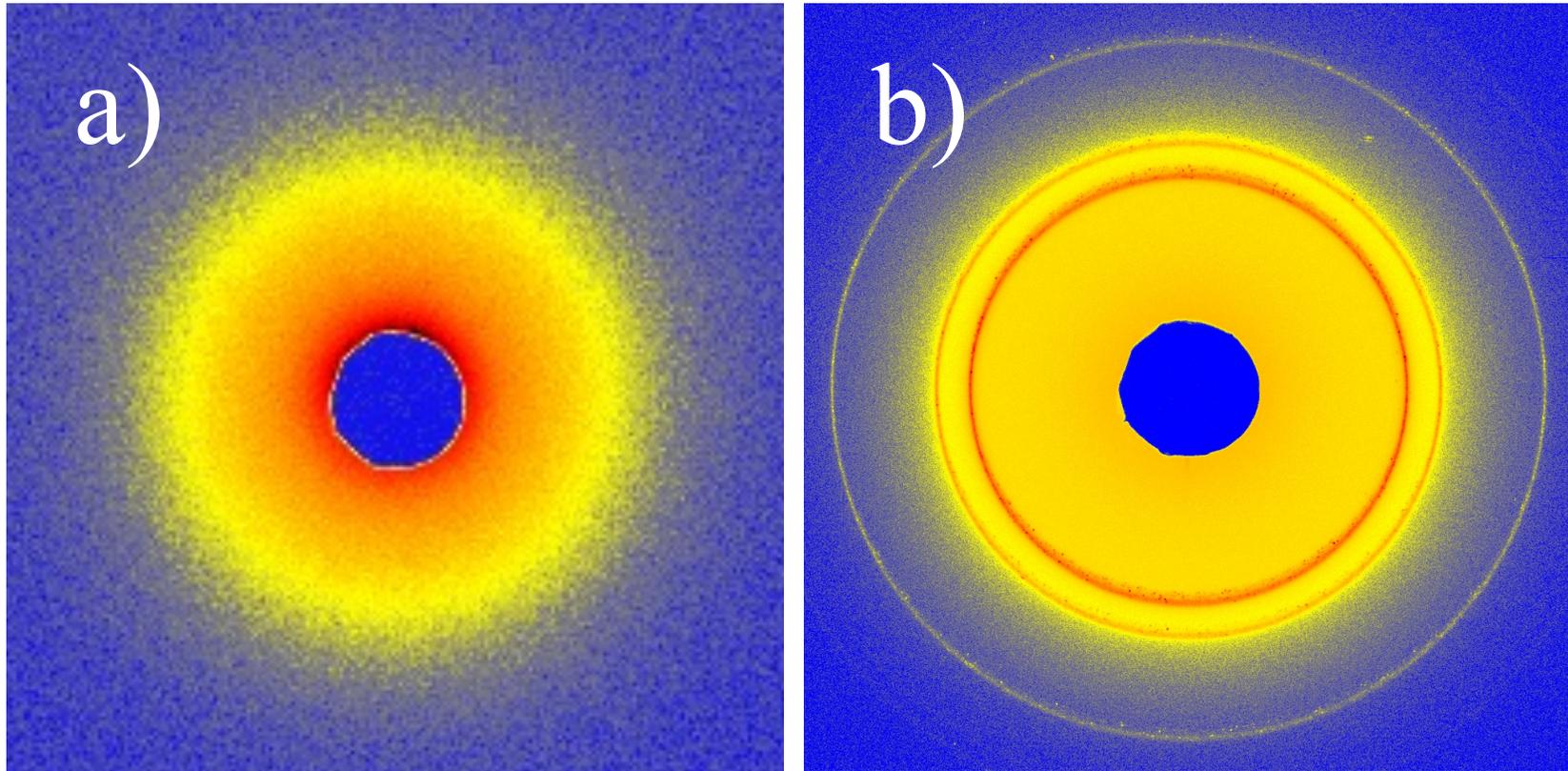
Yau S.-T. and Vekilov P.G., *J. Am. Chem. Soc.* (2001), 123, 1080-1089

- a) Nucleation via planar 2D clustering
- b) Nucleation via compact close-packed 3D clustering
- Possible also via non-crystalline (eg icosahedral) nuclei.



**Figure 10.** Schematic illustration of two nucleation pathways: (a) via a planar critical cluster (in IV molecules belonging to the second layer are shown in a lighter shade; in V the (110) layers that stack up to form this crystal are delineated by lighter and darker contours) and (b) via compact critical cluster.<sup>5,6</sup>

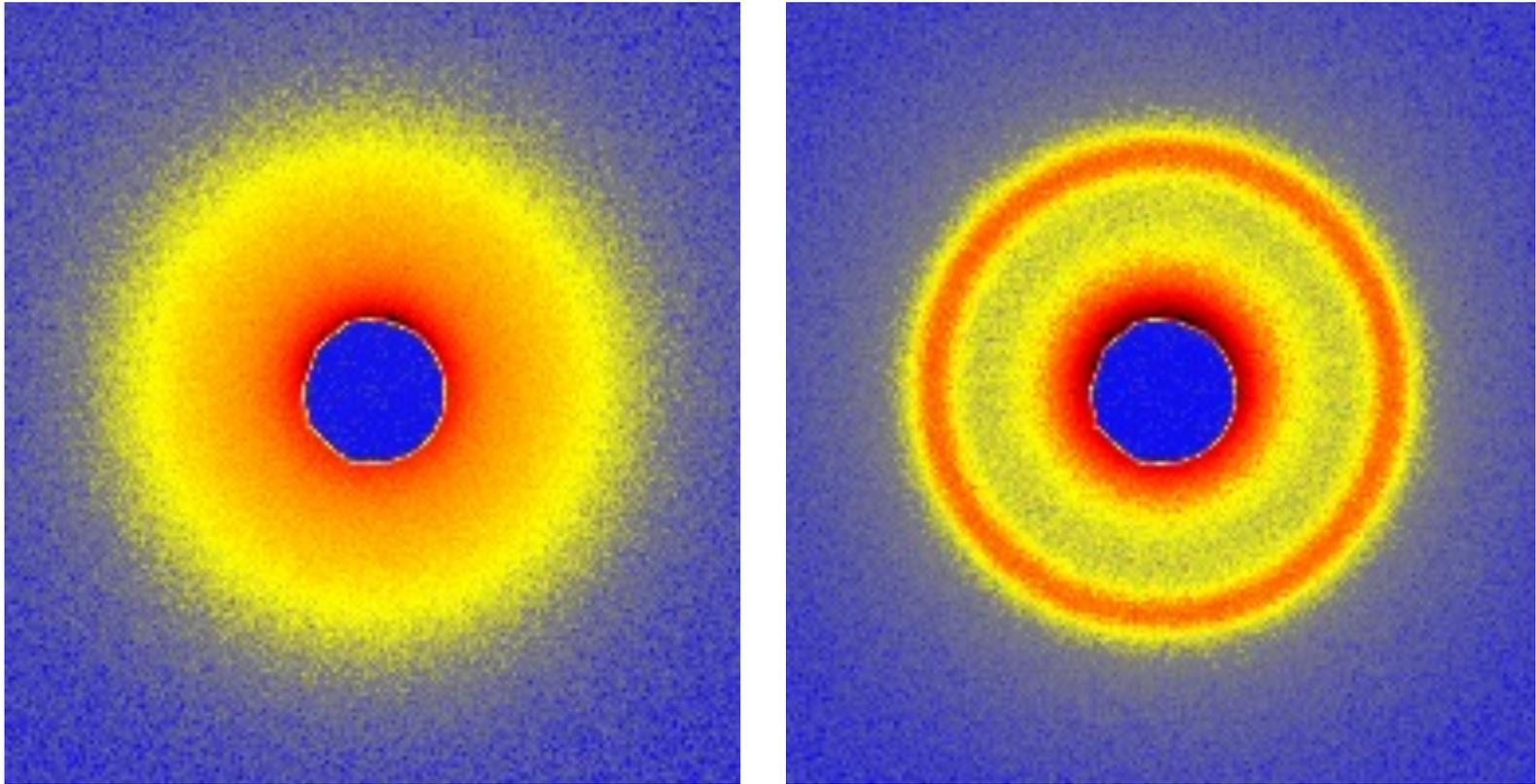
# Crystallization of Holoferitin: SAXS



a) SAXS pattern of holoferitin at 10°C.

b) SAXS pattern after adding Cadmium salt

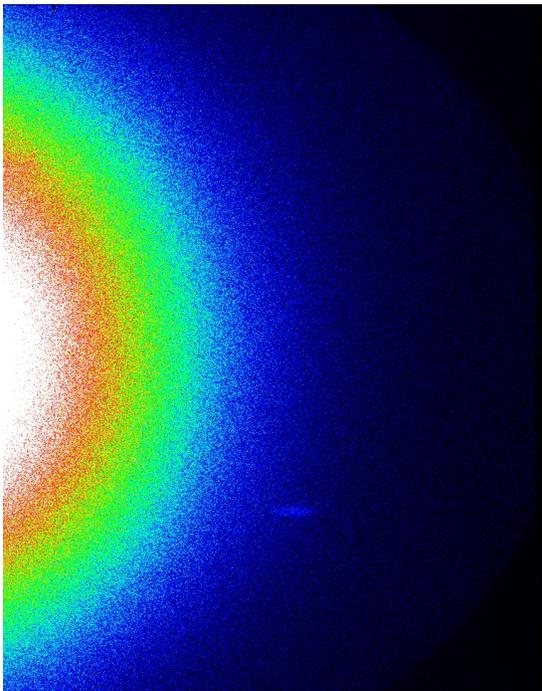
# Effect of Temperature



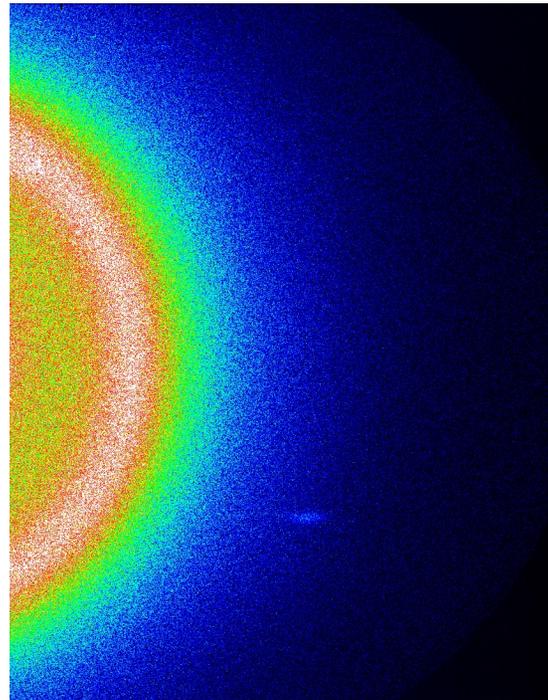
SAXS pattern at  $T=+10^{\circ}\text{C}$  and  $-10^{\circ}\text{C}$ .

Holoferitin in dilute (10mg/ml) NaCl only.

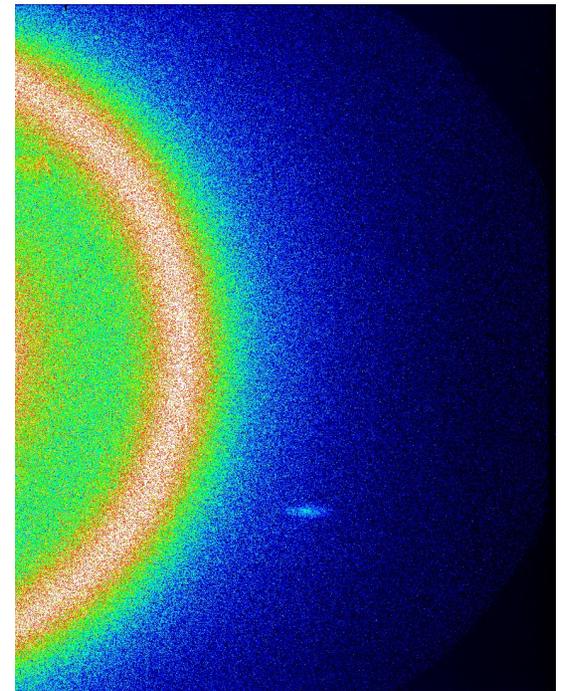
# Ferritin Solution upon Freezing



T=13.3

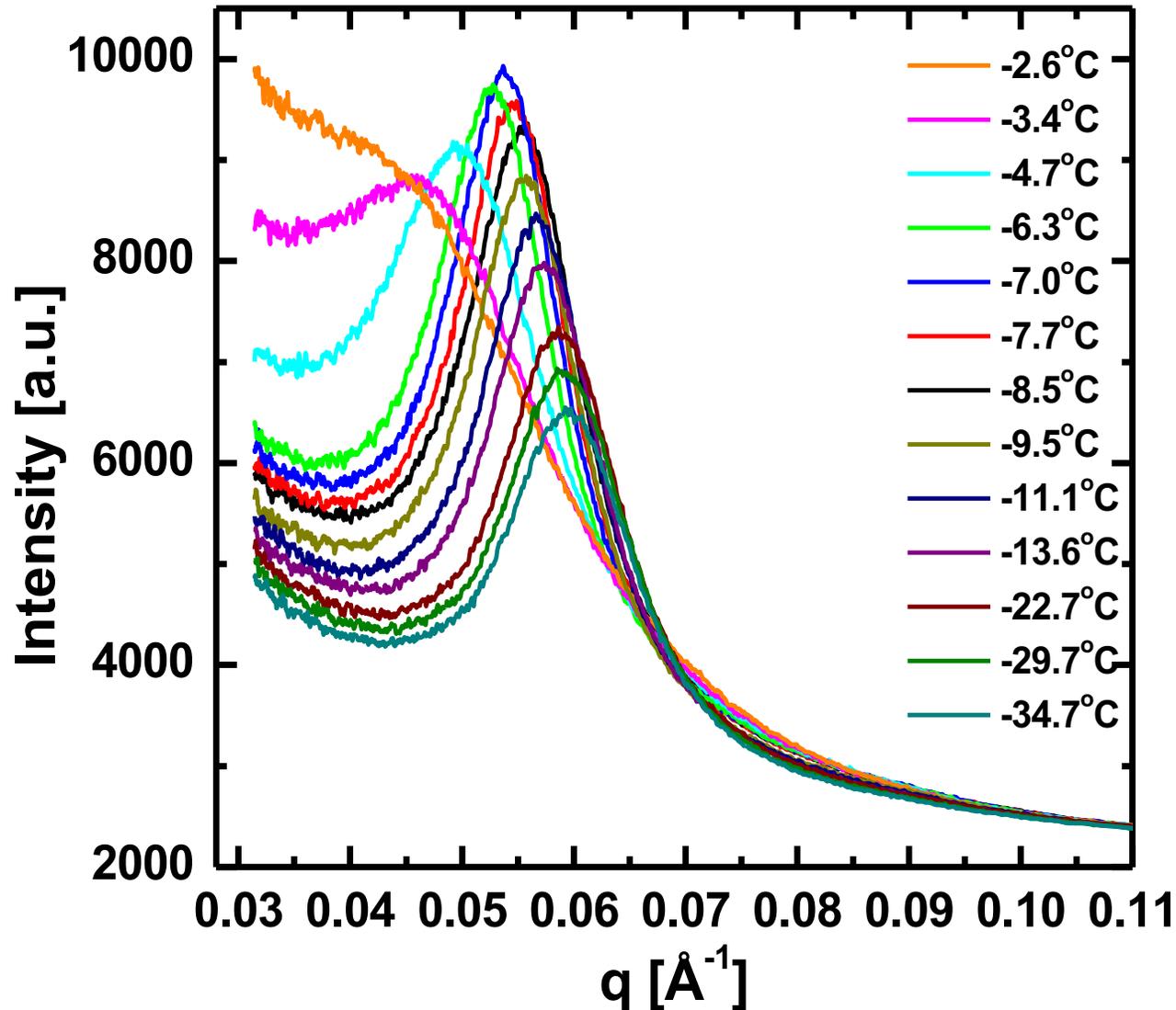


T=-4.7



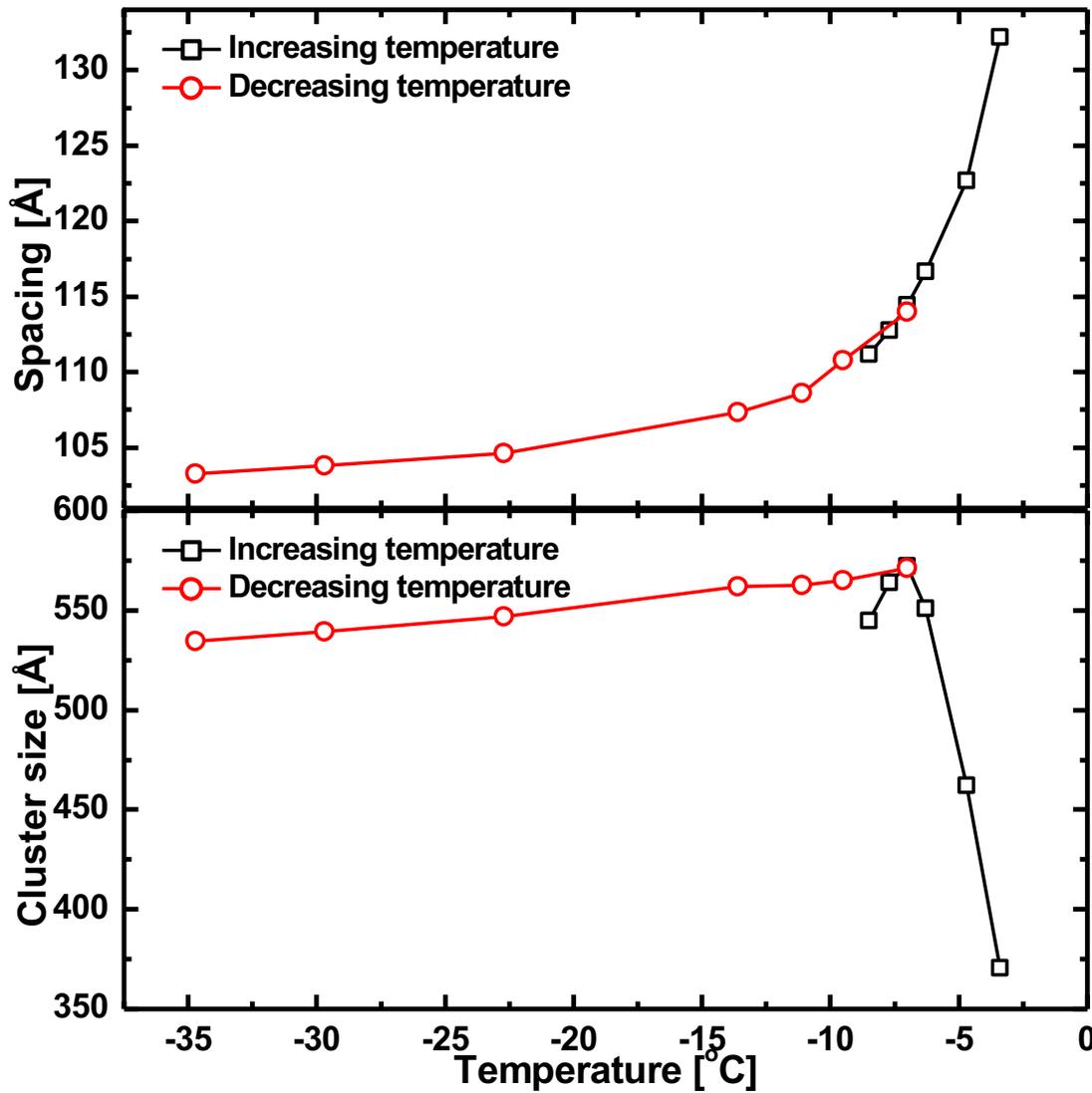
T=-34.7

# Integrated SAXS Data



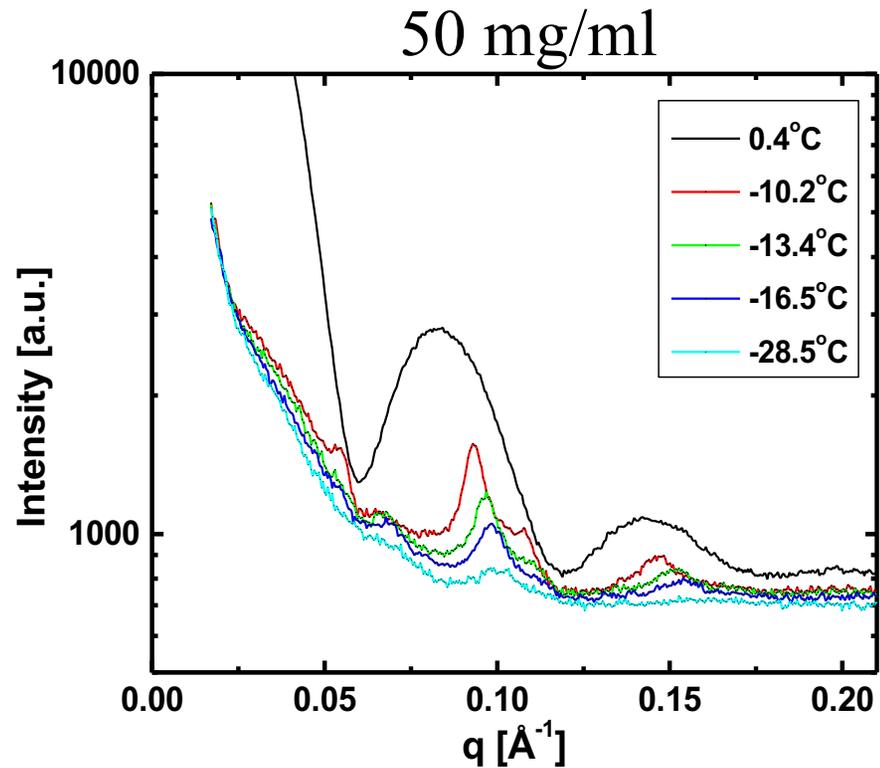
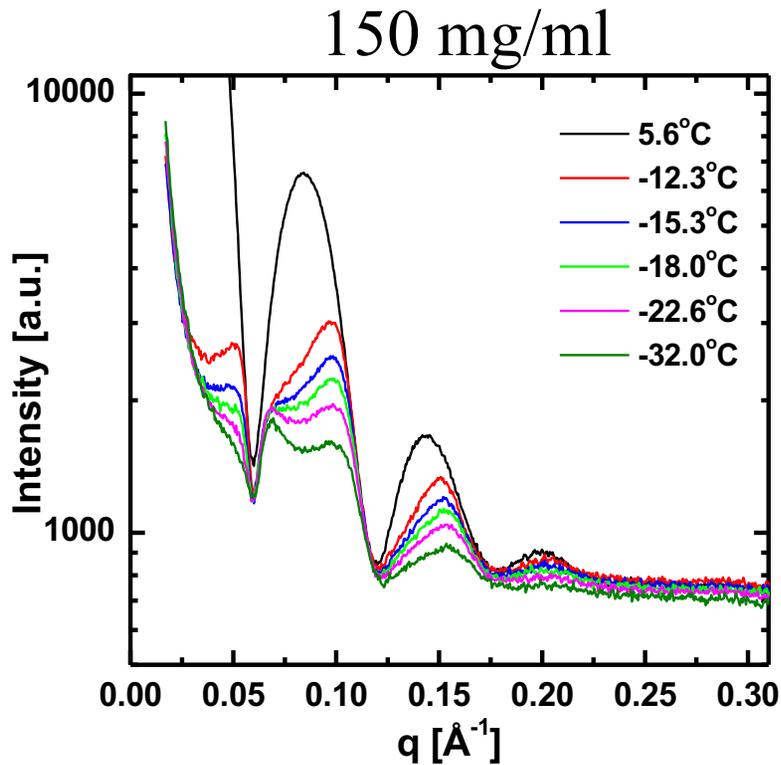
- The SAXS pattern shows a broad peak which shifts in position and changes in width as a function of temperature

# Thermal “Expansion”



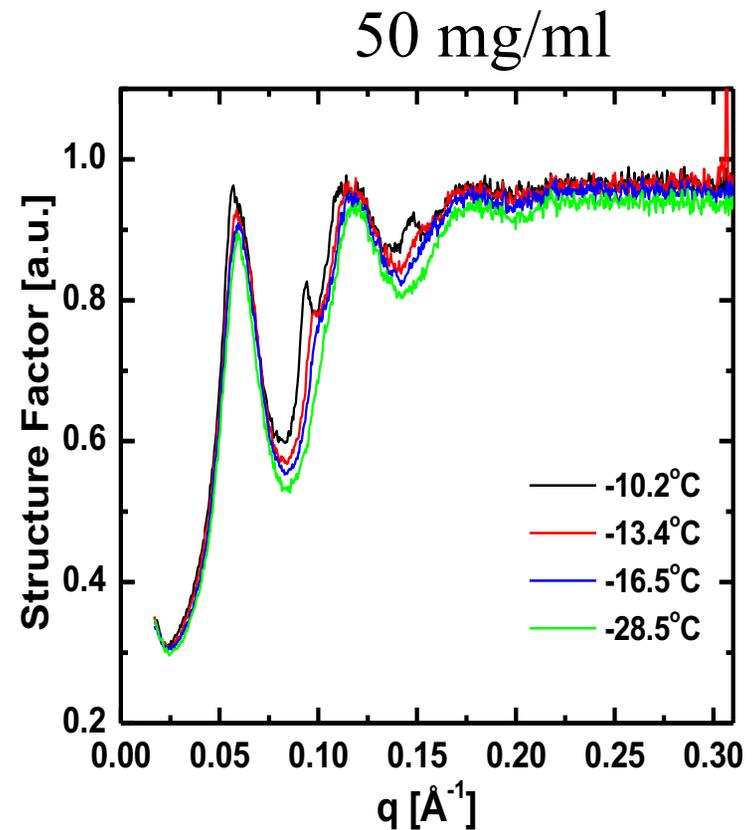
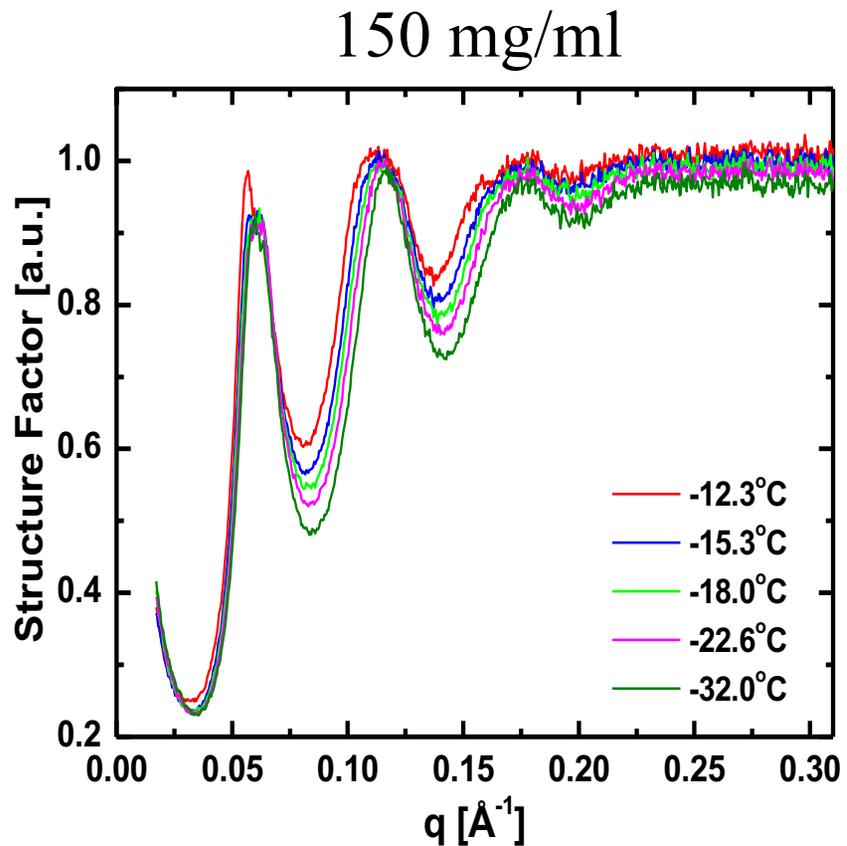
- The position and width of the peaks were fitted with Gaussians.
- The solution freezes and melts at different temperatures.
- The spacing of the molecules in the clusters varies by 30% over 30°C

# Apoferritin Data



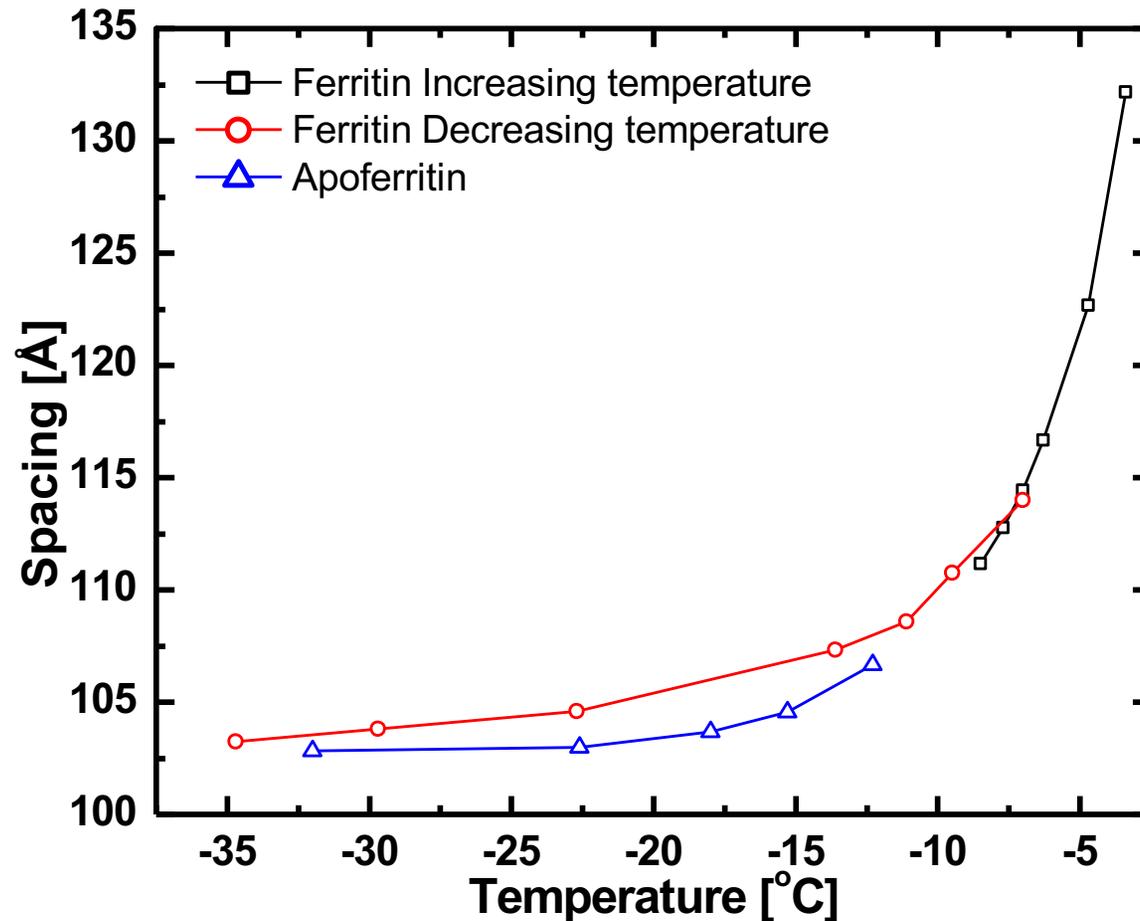
- The SAXS pattern of apoferritin changes drastically when frozen, from the Bessel function like curve to sharper, but weaker peaks.

# Apoferritin Structure Factors



Thermal “expansion” is still present  
Form factor (Bessel function) divided out

# Apoferritin Thermal Expansion



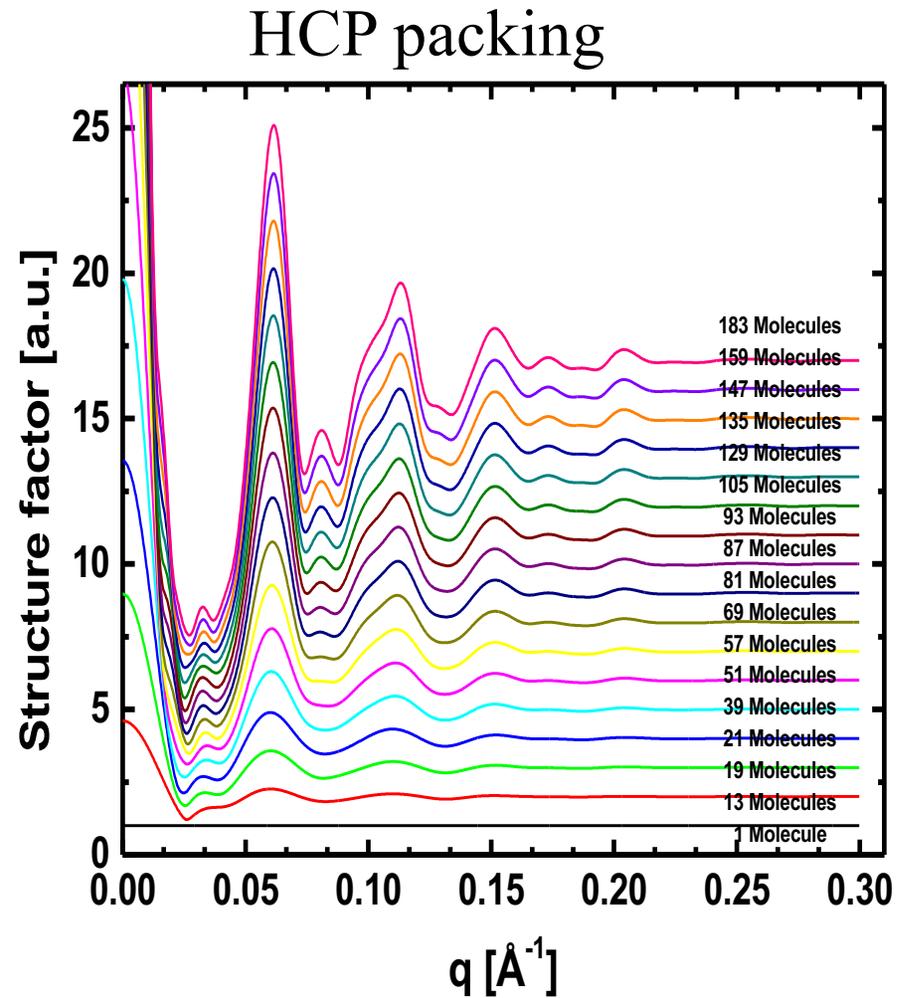
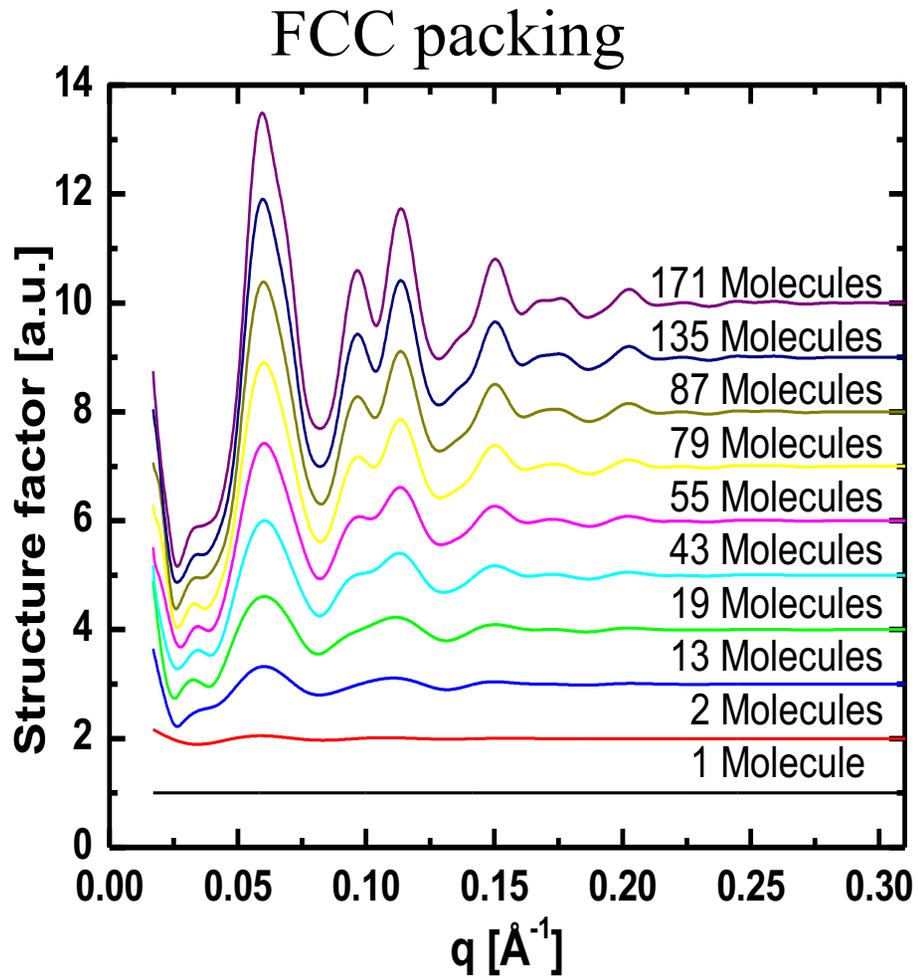
- The thermal expansion of apoferritin matches that measured from holo-ferritin.

# Cluster Simulations

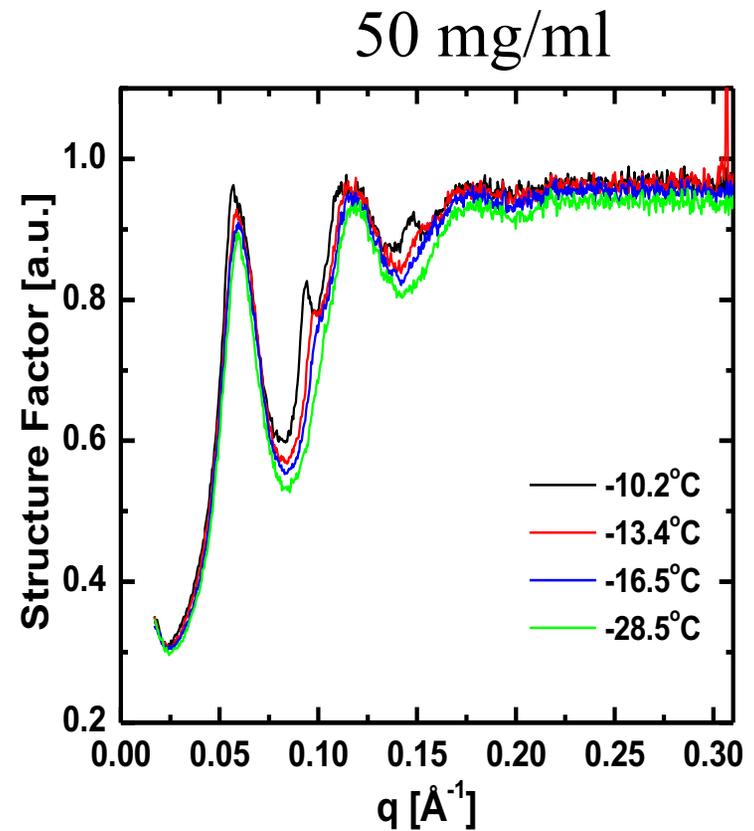
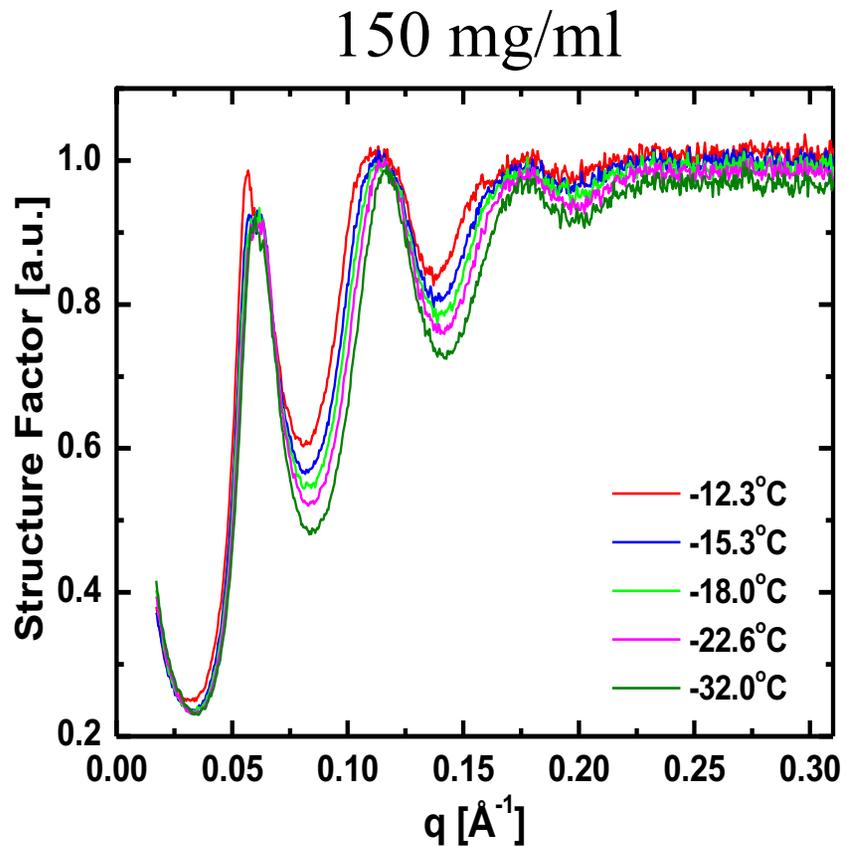
- We used the Debye formula to calculate the spherically averaged structure factor (S) of apoferritin clusters.
- Assume bulk crystal spacing.
- We add protein ‘atoms’ in symmetrical shells around a central molecule and calculate S:

$$S^2(q) = \sum_i^N \sum_j^N F_i F_j \frac{\sin qr_{ij}}{qr_{ij}}$$

# Close-Packed Clusters



# Apoferritin Structure Factors

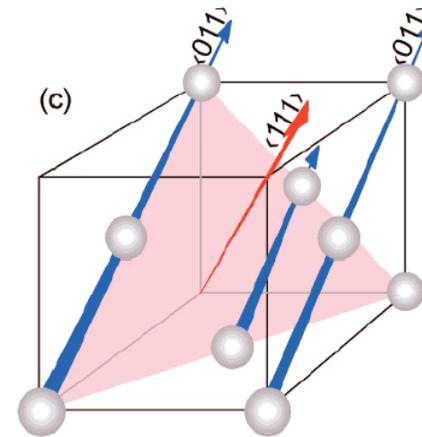
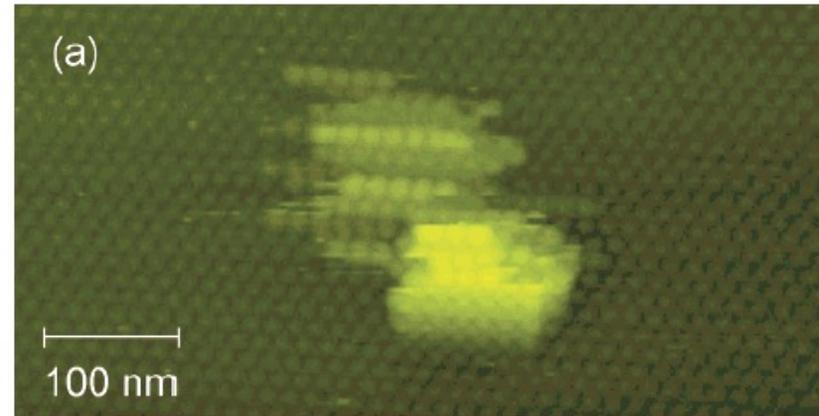


Thermal “expansion” is still present

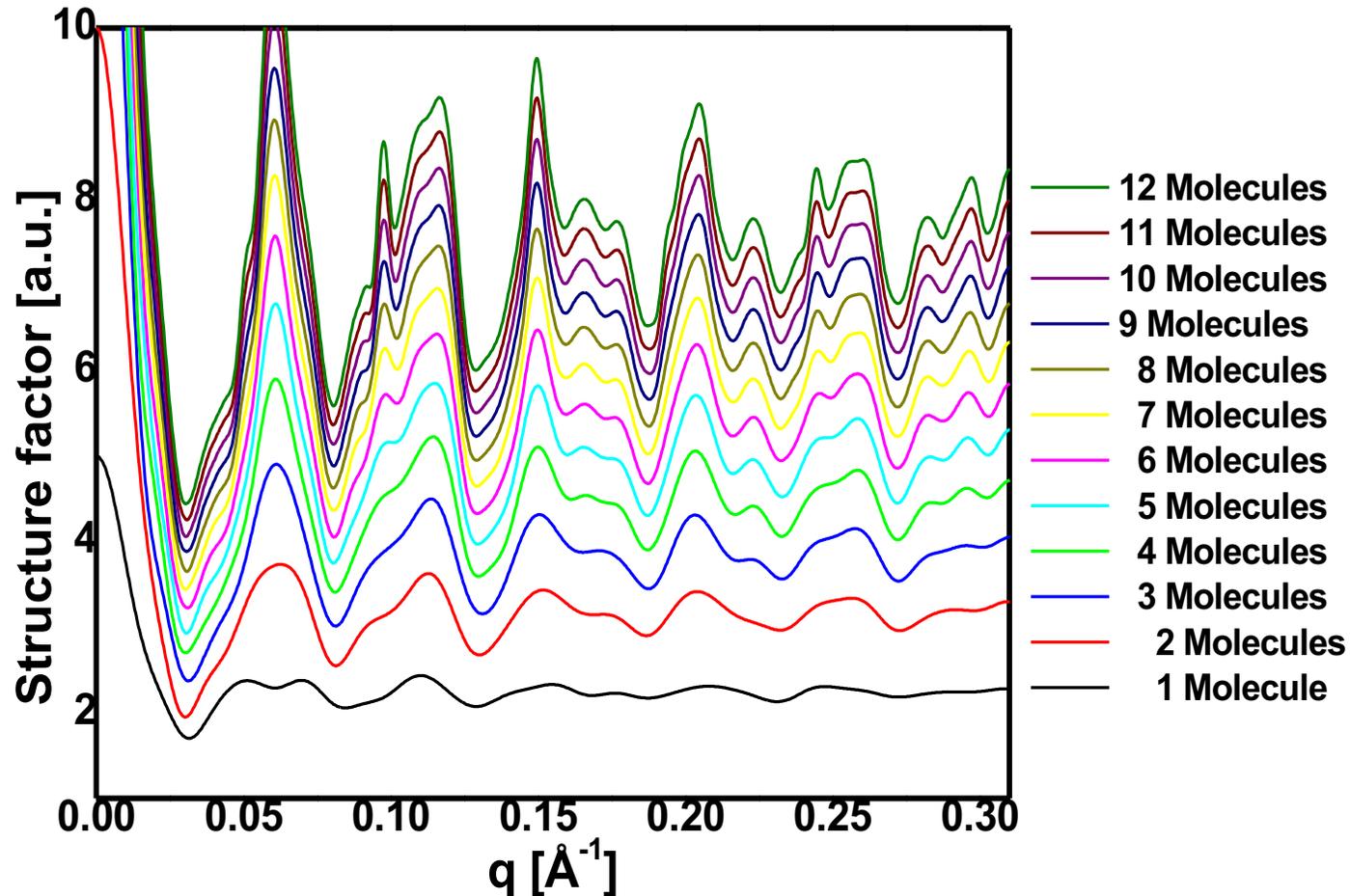
# Planar Ferritin Nucleation

Yau S.-T. and Vekilov P.G., *J. Am. Chem. Soc.* (2001), 123, 1080-1089

- Yau and Vekilov observed by AFM the formation of planar clusters of ferritin made of 4-7  $\langle 110 \rangle$  rods of 4-8 molecules.

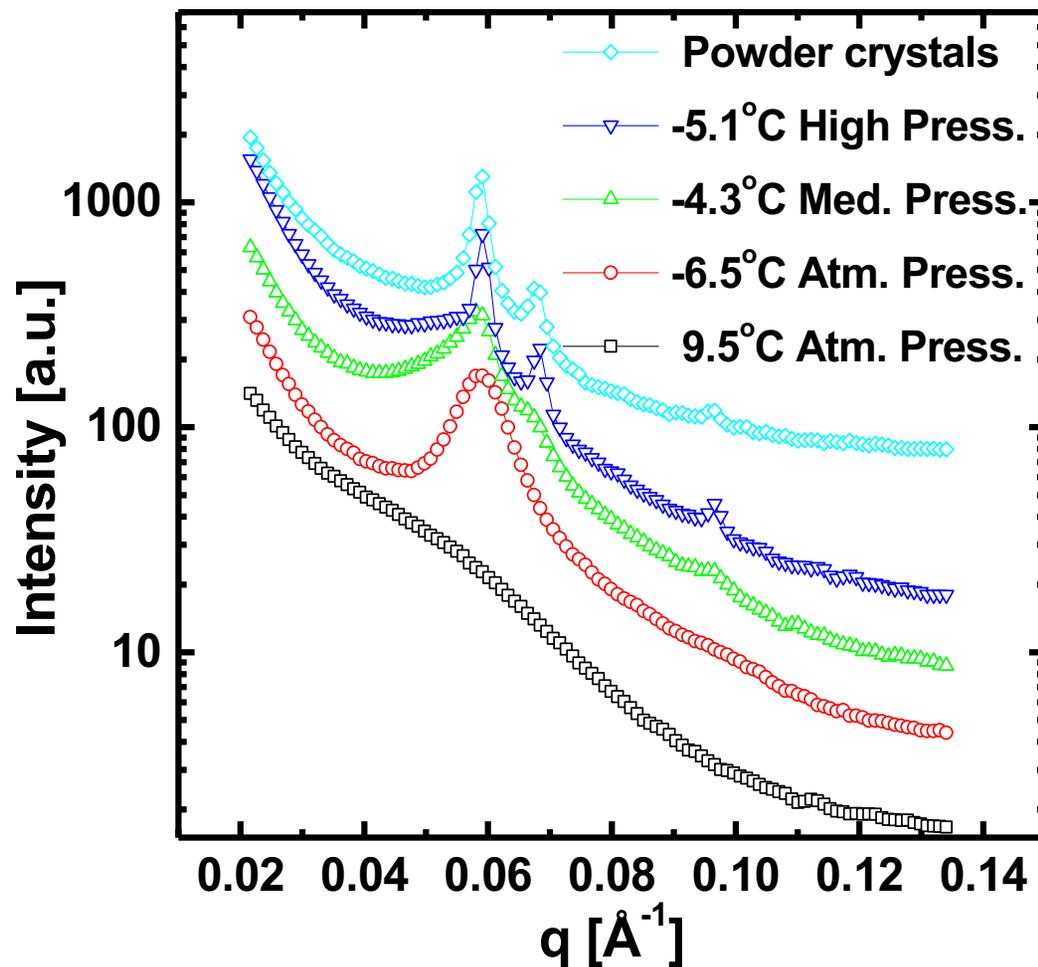


# Planar clusters Simulation



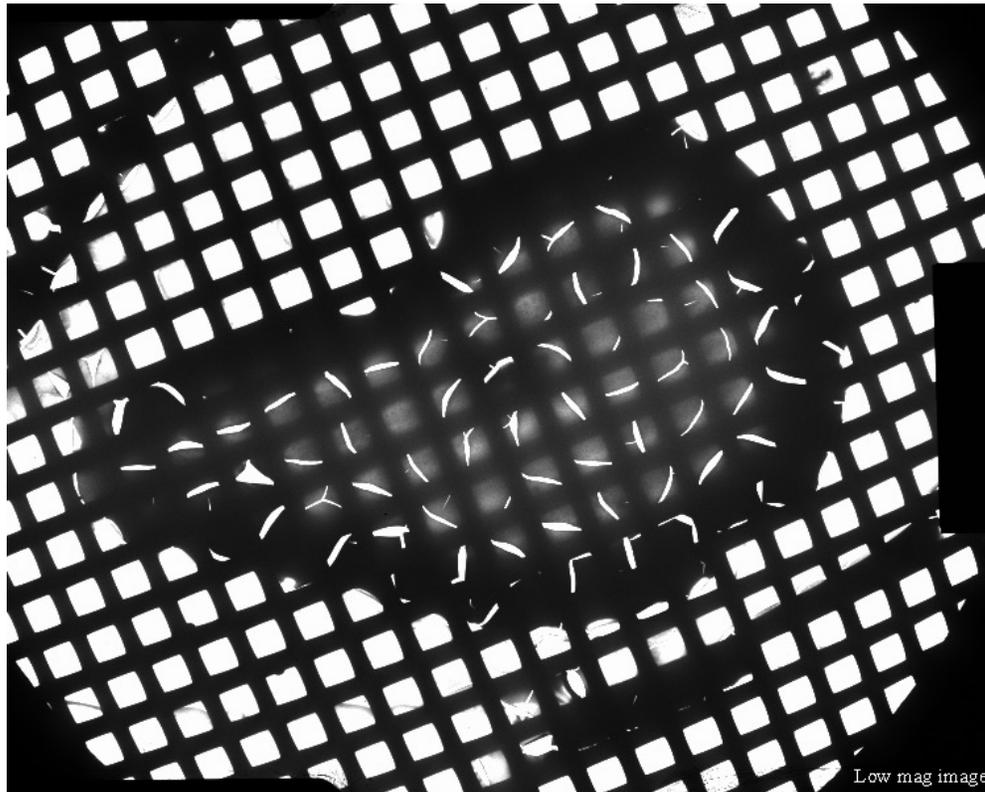
- Simulation of structure factor from clusters of 5  $\langle 110 \rangle$  rods of 1 to 12 molecules per rod

# Hydrostatic Pressure



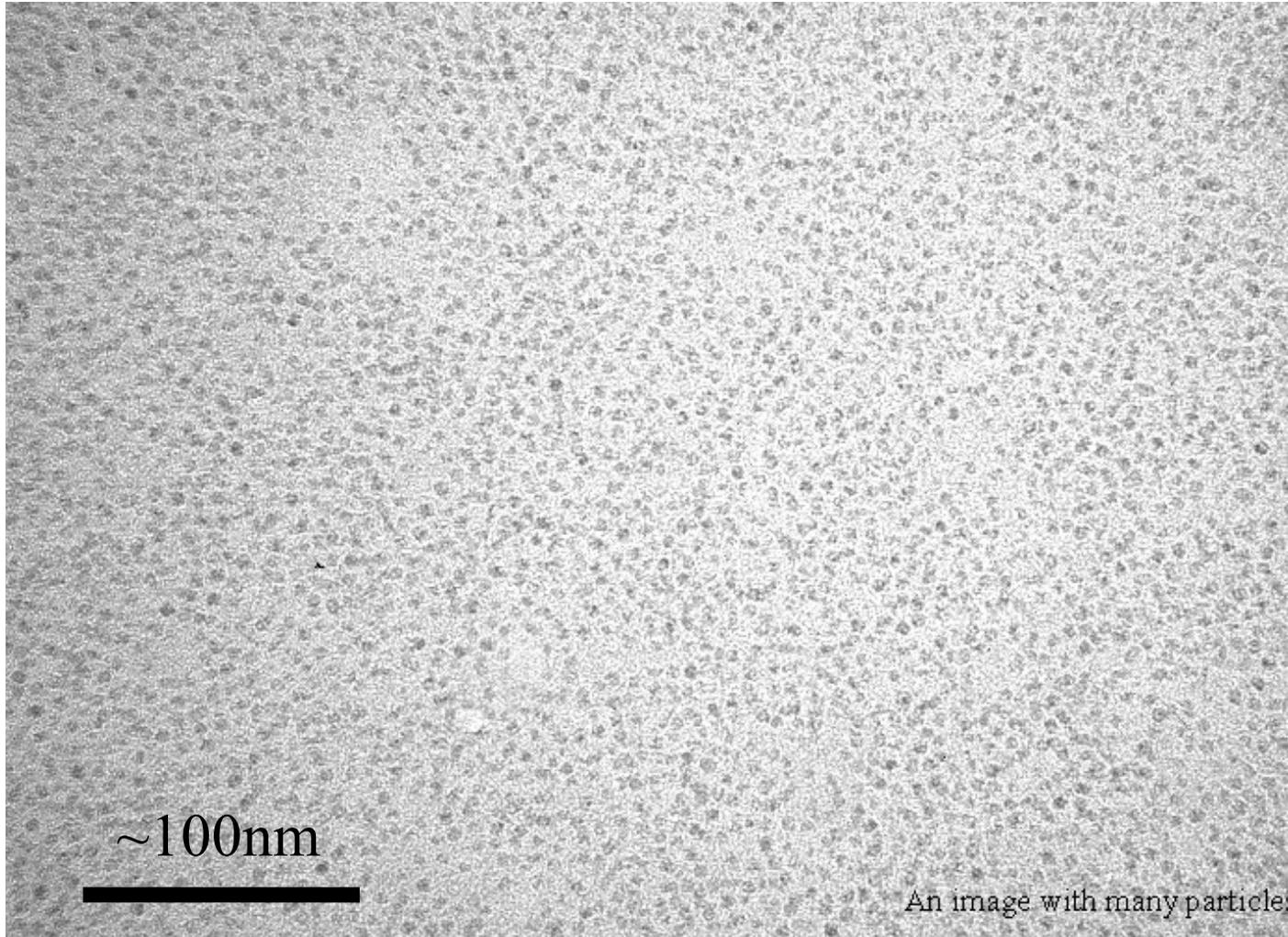
- Pressure generator glued to a sample capillary.
- Pressure transforms the broad peak into sharp Bragg peaks similar to those produced with Cadmium salt.

# Overview of EM grid

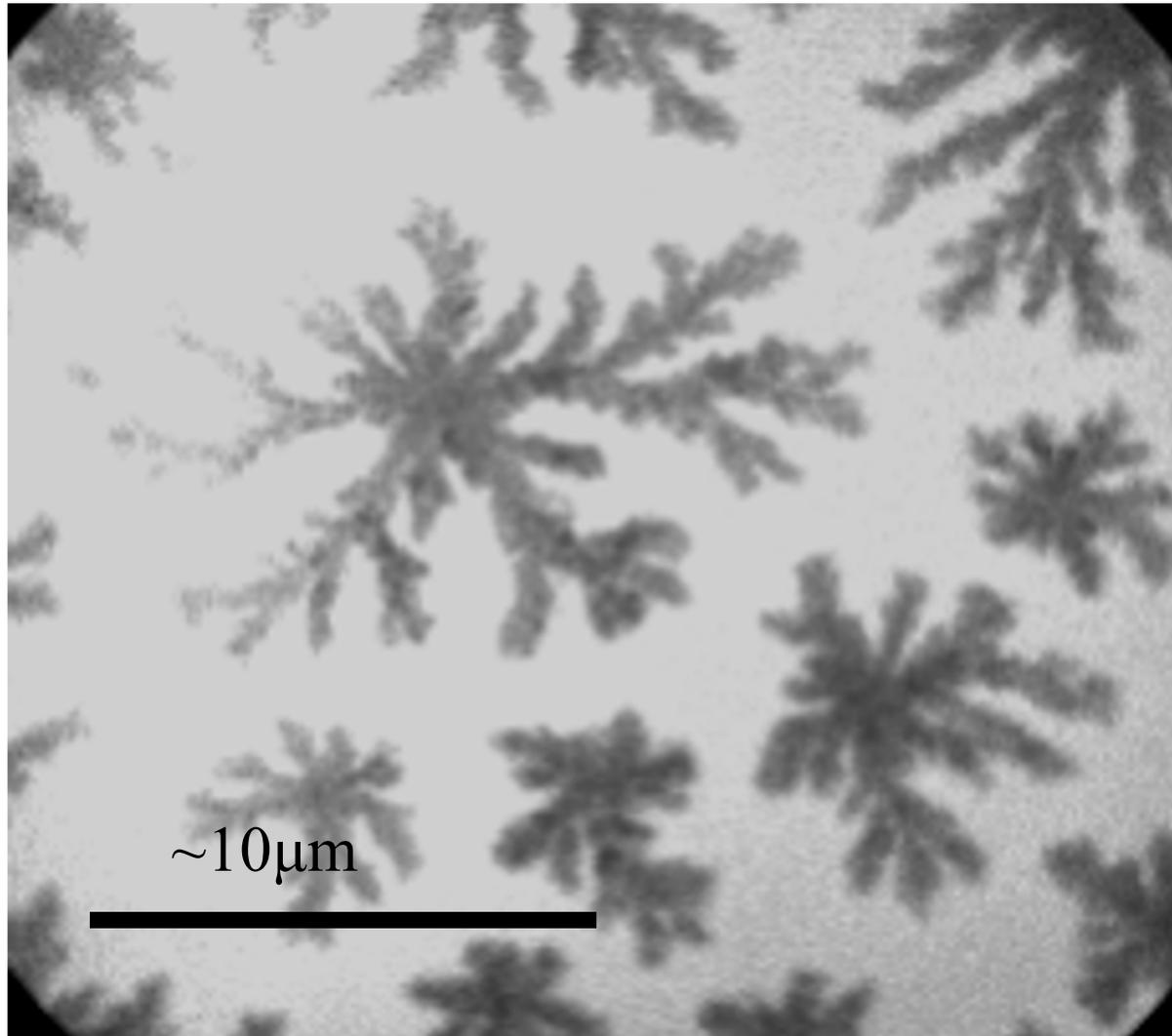


- 3mm diam grid
- a-carbon + “Formvar”
- 10mg/ml Ferritin
- 10mg/ml NaCl
- dipped grid
- dried in air
- no cryo-cooling
  - ▷ three regions of different density

# Border Region (low density)

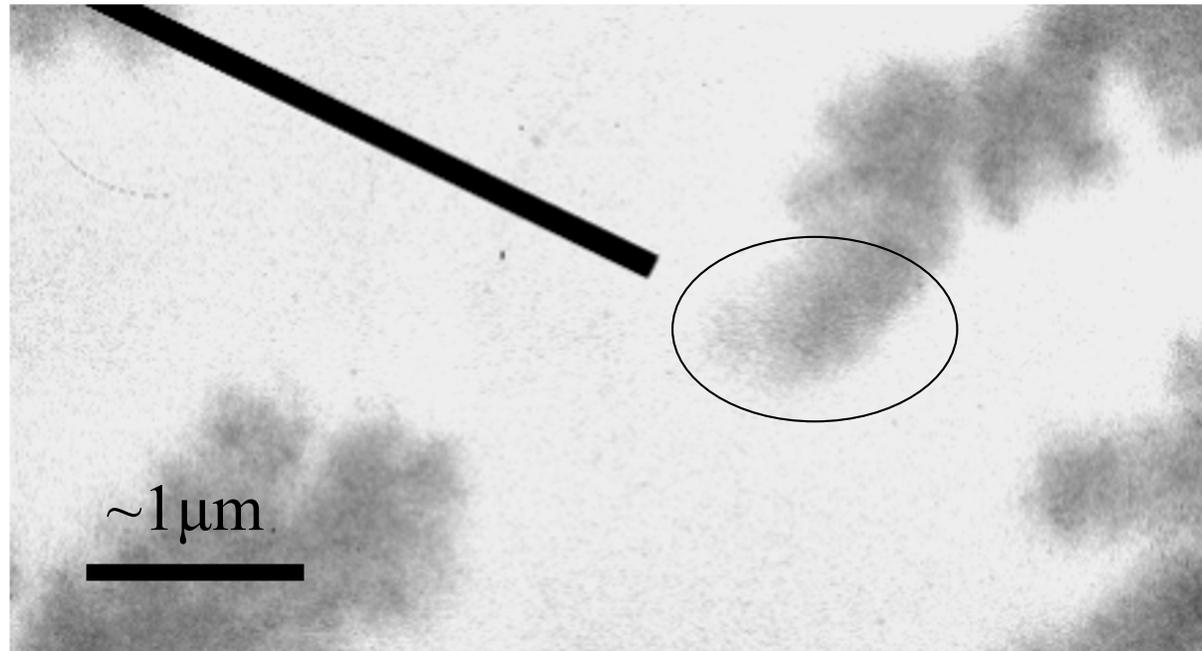


# High Density Ring

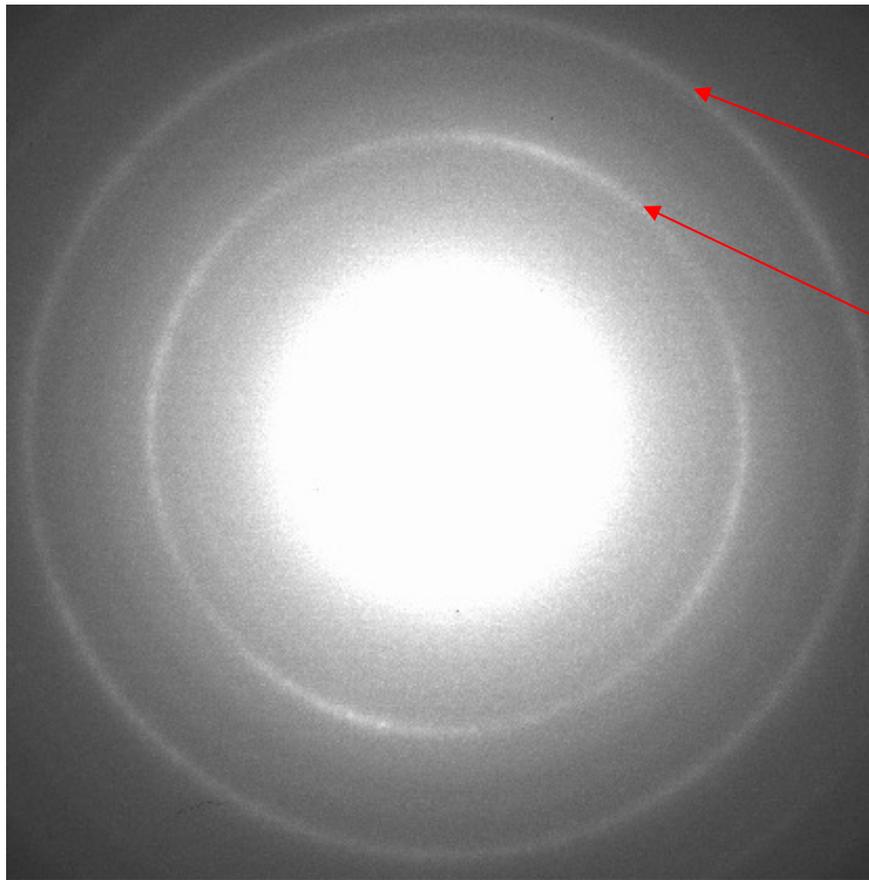


I. K. Robinson Emerging Crystallography

# Select Area around Dendrimer Tip



# Selective Area Diffraction



$$Q=6.32\text{\AA}^{-1}$$

$$Q=4.44\text{\AA}^{-1}$$

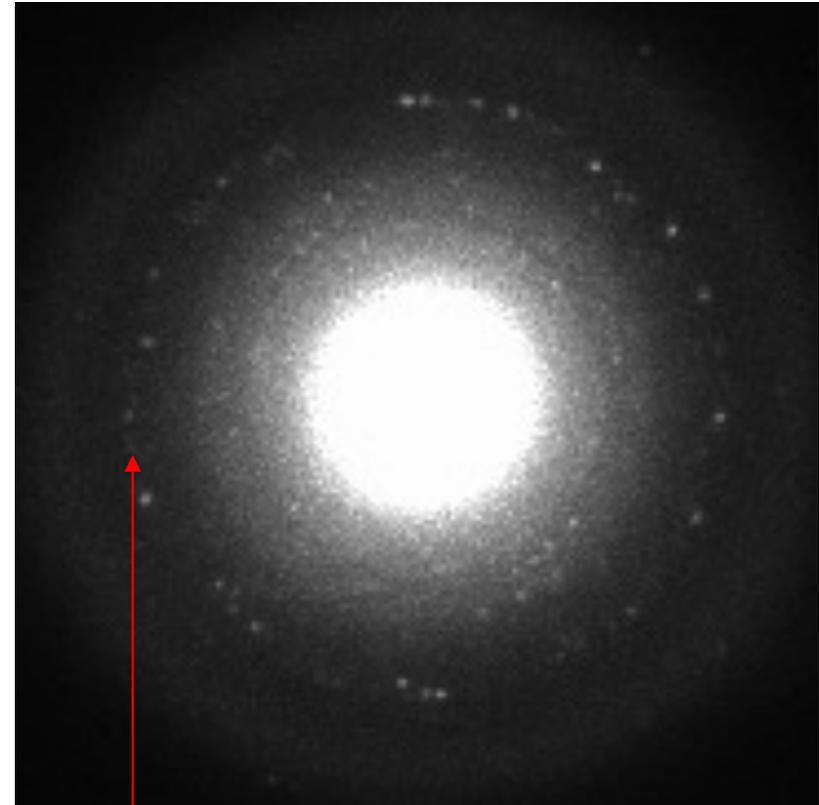
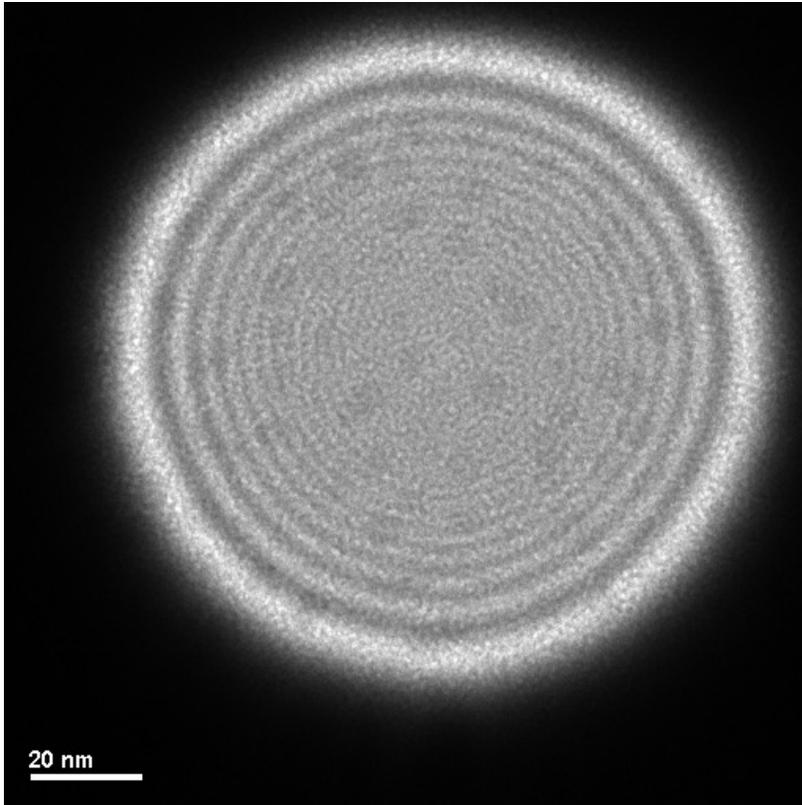
$$\lambda = 0.025\text{\AA}$$

$$D = 2.0\text{m}$$

$$R_1=35.4\text{mm}$$

$$R_2=50.4\text{mm}$$

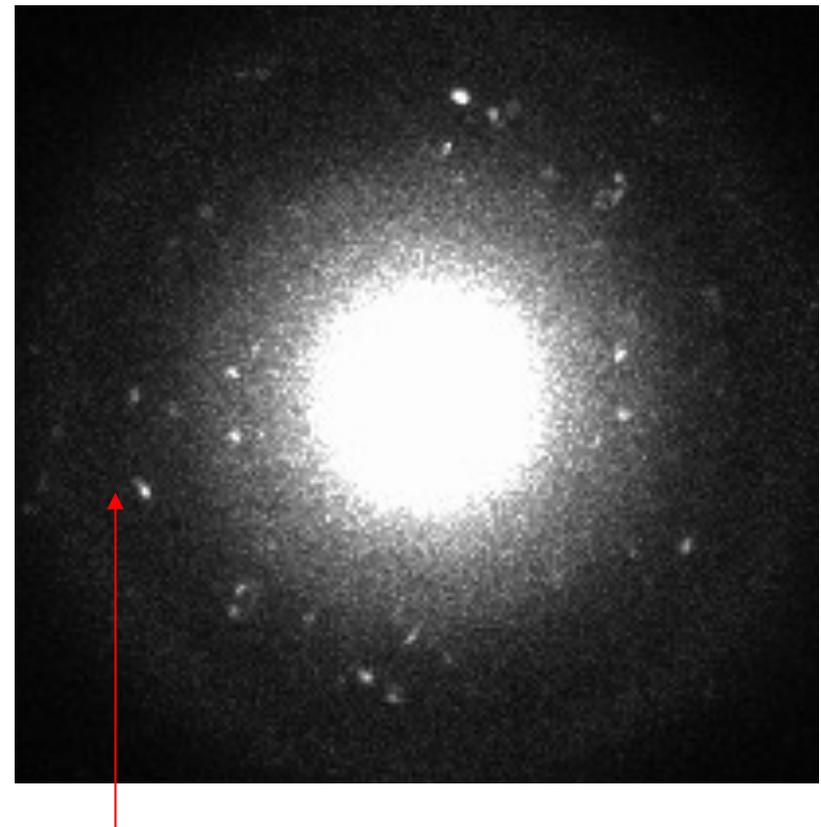
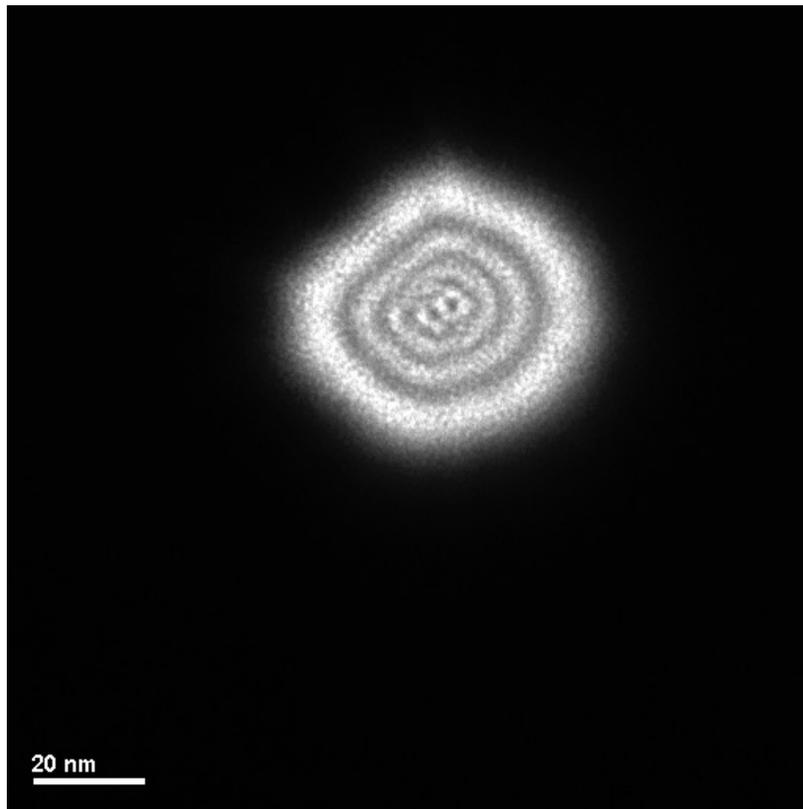
# Diffraction from Border Region



$$Q=8.43\text{\AA}^{-1}$$

# Smaller Area Diffraction

Images: picture2 and diffraction4



I. K. Robinson Emerging Crystallography  $Q=8.2\text{\AA}^{-1}$

# Conclusions

- Smallest Ferritin crystals about 3 microns
- Implosion upon radiation damage
- New condensed state upon freezing
- SAXS consistent with planar or 3D cluster
- Large thermal expansion of cluster
- Single molecule diffraction ideas can be tested by electron diffraction