

Supporting Information

Multimodal Imaging of Autofluorescent Sites Reveals Varied Chemical Speciation in SSZ-13 Crystals

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Supplementary Material

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27 **1 Method**

28 1.1 Sample Preparation

29 SSZ-13 zeolite was synthesized as described previously^[1] using trimethyl adamantammonium
30 hydroxide (Sachem Inc.) as structure-directing agent. Identity and purity of the zeolite were verified by
31 XRD (see below).

32 Calcination in all cases involved placing the sample in a ceramic dish in a static oven and heating in air
33 to 120 °C at 1 °C/min and holding at temperature for 2.5 h, then heating to the target temperature (i.e.
34 550 or 630 °C) at 4 °C/min and holding at the nominated temperature for 10 h.

35

36 1.2 Scanning Electron Microscopy (SEM)

37 SEM images were obtained on a JSM-6701F Field Emission Scanning Electron Microscope (JEOL,
38 Tokyo) using the equipped secondary electron (SE) detector. Images were obtained under high vacuum
39 (5.15×10^{-7} torr/ 6.87×10^{-5} Pa) at 10 kV with scan speed of 735. All samples were coated with gold for
40 35 s, but actual thickness is unknown.

41

42 1.3 Thermogravimetric Analysis (TGA)

43 N₂ flowed at 40 mL/min was used as a balance gas and 60 mL/min of 21% O₂ in He was used during
44 the actual measurement. The samples were heated in air at 1 °C/minute up to 150°C and held for 2.5
45 hours. They were then heated to 350 °C at 2.2 °C/minute and held for 3 hours, and then heated at
46 0.8 °C/min to 580 °C and held for 3 hours.

47

48 1.4 X-ray Diffraction (XRD)

49 Powder X-ray diffraction was measured using a Rigaku Miniflex X-ray instrument with a Cu tube
50 source of 600 mW and 1D D/teX detector and divergent slits. Diffraction patterns were collected
51 between 5-50 2 θ with an increment of 0.017 (2 θ) and an acquisition time of 1 sec/step.

52

53 1.5 Confocal Fluorescence Microscopy

54 Confocal images were obtained on a Leica SP8 microscope (Leica Microsystems, Germany) at by
55 exciting with a continuous wave diode laser at 405 nm (maximum power 7 mW at the sample) and with
56 a pulsed supercontinuum laser (NKT Photonics, Denmark, maximum power 50 mW at the sample). A
57 63x water-immersion objective with 1.2 numerical aperture (Leica, HC Pl Apo, 63x/1.20) was used to
58 collect images of 1024 x 1024 pixels with a field of view of 50x50 μm^2 . The scanning speed of the
59 galvanometric mirrors was set to 100 Hz, while the pinhole was opened to 1 Airy Unit at 580 nm. Notch
60 filters at 405 nm or 488 nm were used. Single-colour images were collected using a single hybrid
61 detector with a detection window set from 450 to 650 nm. Two-colour images were collected using two
62 hybrid detectors at the same time, the first detection window spanning from 450 to 550 nm and the

63 second detection window spanning from 550 to 650 nm. Z-stacks were collected with a z-step of 0.4 –
64 0.8 μm . Image visualisation and 3D reconstruction of the image stacks was performed in ImageJ (Fiji)^[2].
65

66 1.6 Fluorescence Lifetime Imaging Microscopy (FLIM)

67 FLIM images were obtained on the same Leica SP8 microscope used for confocal imaging. In this case,
68 the samples were excited only at 488 nm with a pulsed supercontinuum laser (NKT Photonics,
69 Denmark) with repetition rate reduced to 10 MHz, which gave a timespan for the lifetime decay of
70 almost 100 ns. The same 63x/1.2 water-immersion objective was used to collect both single plane FLIM
71 images and 3D stacks of FLIM images of 512x512 pixels. The emission was collected from 515 nm to
72 700 nm. Integrated photon counting hybrid detectors were used in conjunction with a PicoHarp 300
73 TCSPC module (PicoQuant) (PicoQuant GmbH, Germany). The acquisition lasted 8 minutes for single
74 image experiments, while in 3D experiments planes were 1 μm apart and the total acquisition time was
75 30 minutes.

76 Analysis of FLIM datasets was performed with the *FLIMfit* software tool (version 5.1.1) developed by
77 Warren et al.^[3]. The images were binned twice in space and in time (4x4 spatial binning for the 3D
78 FLIM). An experimental instrument response function (IRF) was used in the analysis. Phasor plots were
79 generated via *FLIMfit* and used to segment the images on the basis of a common lifetime. The phasor
80 plot method^[4] provides a 2D polar plot representation of the lifetime decay histogram for each pixel,
81 after transforming it into a phasor. A pixel-wise analysis was performed with a tri-exponential model
82 after setting up an intensity threshold of 25 photons and a 5x5 smoothing kernel. Offset and IRF shift
83 were left as parameters of fitting. Areas of similar lifetime were extracted with the phasor approach and
84 then averaged and fitted. The 3D FLIM reconstructions were made with ImageJ (Fiji).

85

86 1.7 Multispectral Imaging

87 Multispectral imaging was performed using the λ -scan feature of the Leica SP8 microscope (Leica
88 Microsystems, Germany) in order to extract spectral information from the confocal images. Multiple
89 images of the same sample position were acquired with fixed excitation, while the detection band, set
90 to 7.5 μm , was tuned by shifting the detection slit of 5 μm . Multispectral images with excitation at 405
91 nm were acquired from 430 nm to 690 nm, while multispectral images run at 488 nm were acquired
92 from 500 nm to 690 nm. Spectra were extracted with the Leica SP8 LASX software by selecting regions
93 of interest.

94

95 1.8 Raman Spectroscopy

96 Raman spectra were obtained from an InVia confocal Raman microscope (Renishaw, Wotton-under-
97 Edge, UK) equipped with a 50x objective lens (Nikon, L Plan Apo, 50x/0.45, WD17) and Peltier cooled
98 CCD. The sample was irradiated with either an 830 nm diode laser with 34 mW maximum power output,
99 or a 514 nm Stellar PRO argon laser (Modu-laser, USA) with 10 mW maximum power output.

100 Attenuated laser power at the sample was achieved using neutral density filters. A grating with 1200
101 lines/mm was used, yielding an overall spectral resolution of 1 cm^{-1} .

102 Raman maps were acquired using an x-y motorised stage, which was moved in $1.5\text{ }\mu\text{m}$ steps to raster
103 across a region of interest. At each point, Raman spectra were collected in the spectral region 1015 to
104 1472 cm^{-1} with an acquisition time of 20 s and 10 accumulations totalling an overall acquisition time of
105 18-20 hrs per map, and an irradiance of 5000 W/cm^2 .

106 The spatial distribution of the most significant Raman bands were reconstructed by calculating the
107 correlation of a triangular template (centered at the frequency ν with a base width $\Delta\nu$ of 20 cm^{-1}) with
108 the Raman spectrum detected in each analysis point in the spectral band $\Delta\nu$. The following method,
109 quoted as TC descriptor^{[5],[6]}, is more selective than other mapping methods^[7].

110 Since 830 nm was used with a long working distance objective lens, the nominal spot dimension is
111 around $5\text{ }\mu\text{m}$ and the depth of field is at least $10\text{ }\mu\text{m}$. The entire z-depth is likely to be excited as the
112 spectra were acquired without a pin hole, relatively high laser power, and the crystal medium's low
113 absorption and tendency to scatter allows for greater photon propagation through the volume.

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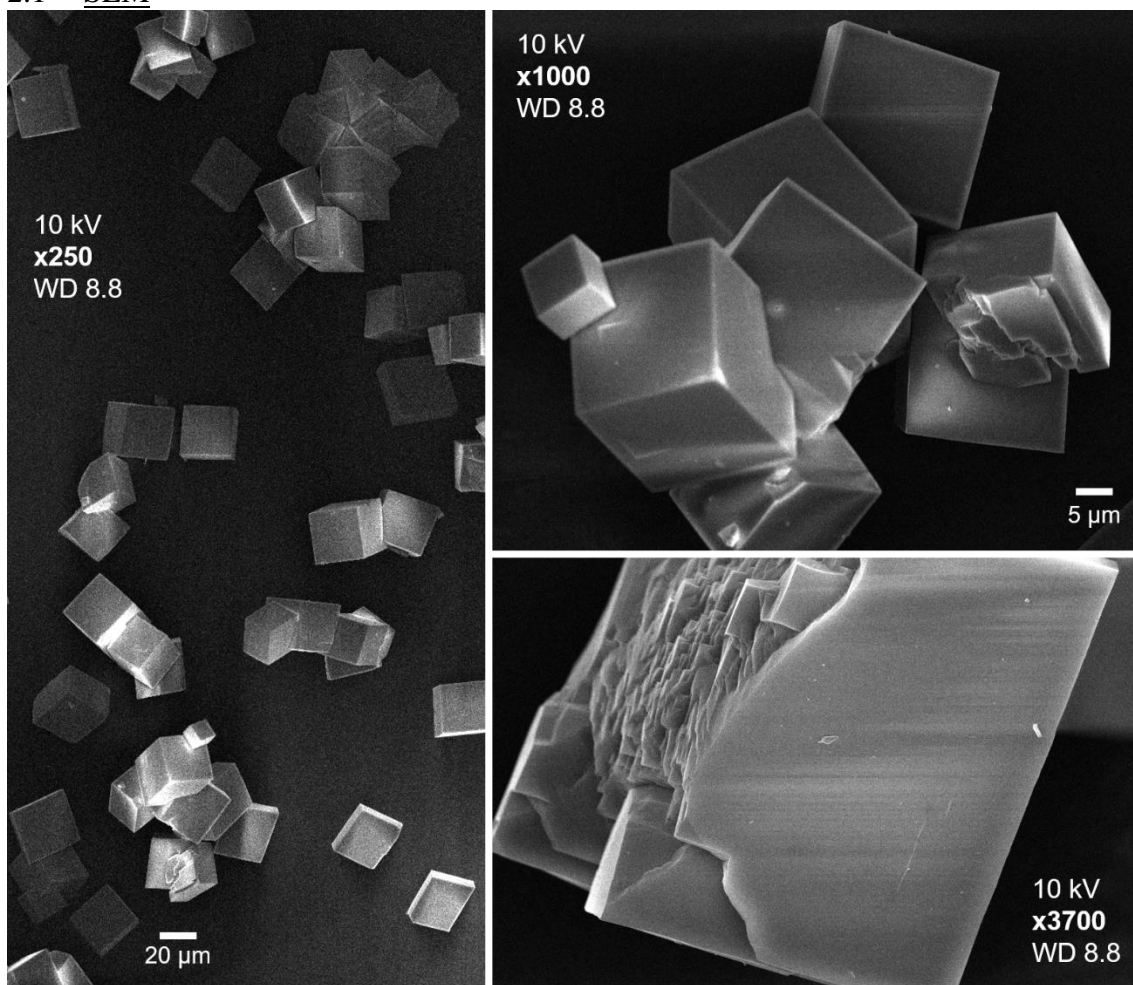
115 1.9 X-ray Fluorescence (XRF) Imaging

116 The X-ray Fluorescence (XRF) experiment was performed at the 2-ID-E microprobe hard X-ray
117 beamline, at the Advanced Photon Source of Argonne National Laboratory, using 10keV X-ray photons,
118 selected by a Si(111) single crystal monochromator^[8]. The X-ray beam was collimated by a $500\text{ }\mu\text{m}$
119 diameter pinhole and focused by a Fresnel Zone Plate (FZP) onto the sample. An order-sorting aperture
120 and central stop were used to filter out the unfocused beam and higher order diffraction. The beam size
121 was $\sim 700\text{nm}$, which defines the spatial resolution. The XRF signals were collected by a Si-drift
122 Vortex®-ME4 energy-dispersive detector. A $30\text{ }\mu\text{m}$ SSZ-13 crystal, charged with copper using "wet
123 ion exchange" once in Cu^{2+} , was raster-scanned at 111 projections angles over a 330° angular range in
124 3° increments. The projective XRF data were processed by MAPS^[9] to deconvolute single element maps.
125 These maps were then aligned and reconstructed by MAPStoTomoPy^[10] to 3D elemental distribution,
126 using a penalized maximum likelihood algorithm in 100 iterations.

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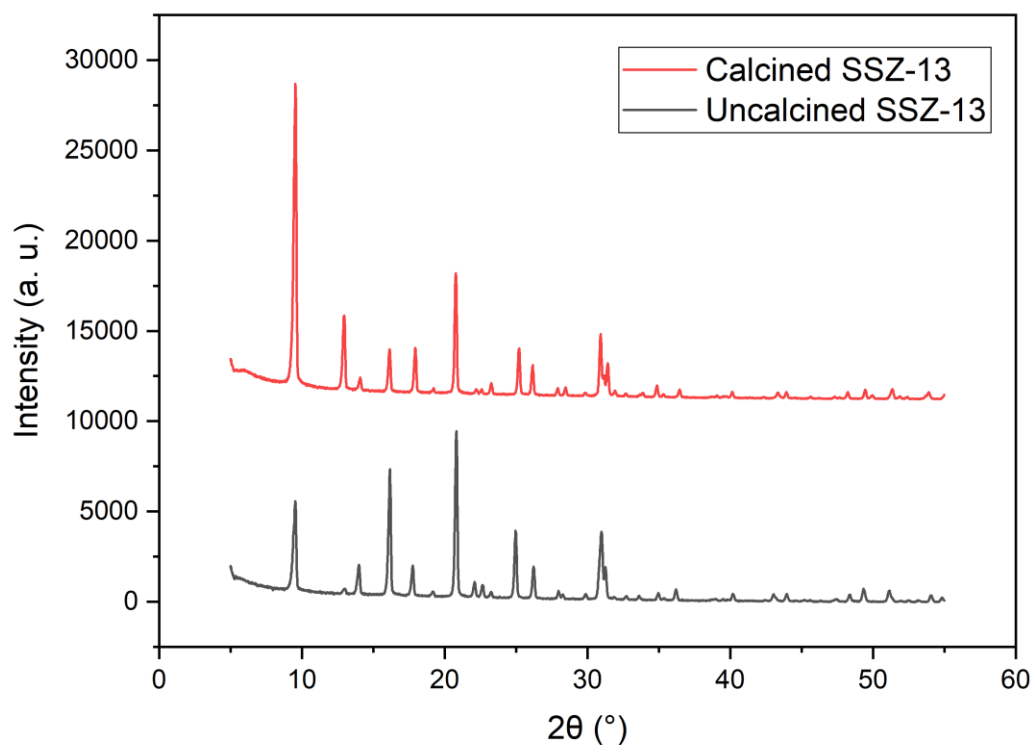
128 **2 Additional Data**

129 **2.1 SEM**



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131 *Figure S 1 SEM images of chabazite crystals calcined at 550°C including widefield image (left), x1000 magnification*
132 *cluster view and x3700 single crystal zoom (right)*

133 SEM images show crystals have a regular cubic crystal morphology. Some evidence of cracking is
134 seen in specific crystals.



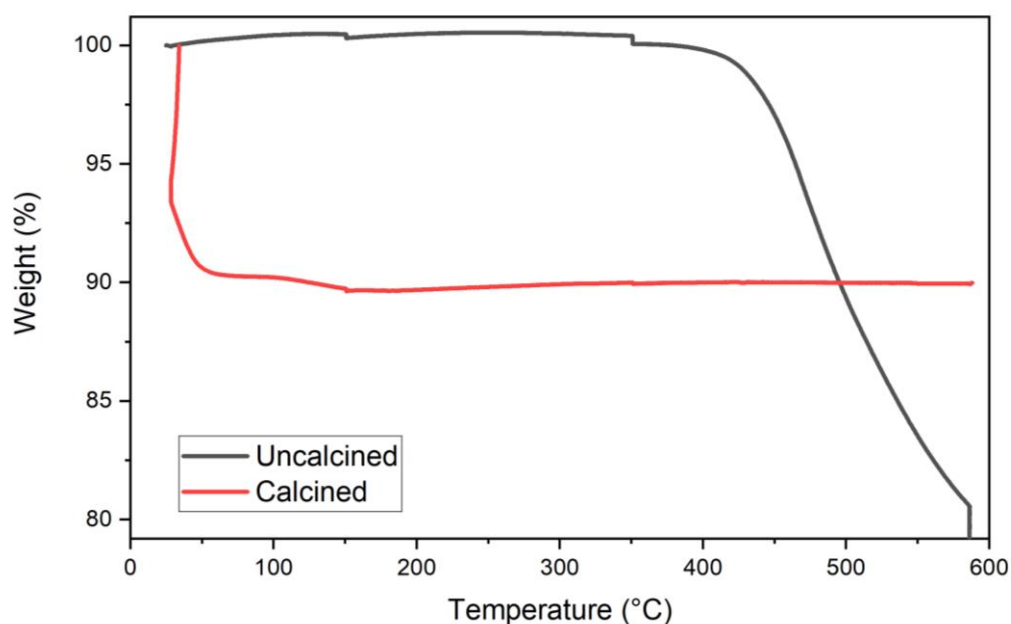
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137 *Figure S 2 XRD patterns for calcined (red) and uncalcined (black) SSZ-13 material*

138 Figure S2 shows XRD patterns for the uncalcined and calcined SSZ-13 material. The clear ensemble of
139 peaks in both samples indicates the satisfactory crystallinity of both samples, further implying that the
140 framework has not been adversely affected by the thermal calcination process. The position of peaks is
141 consistent with IZA data^[11].

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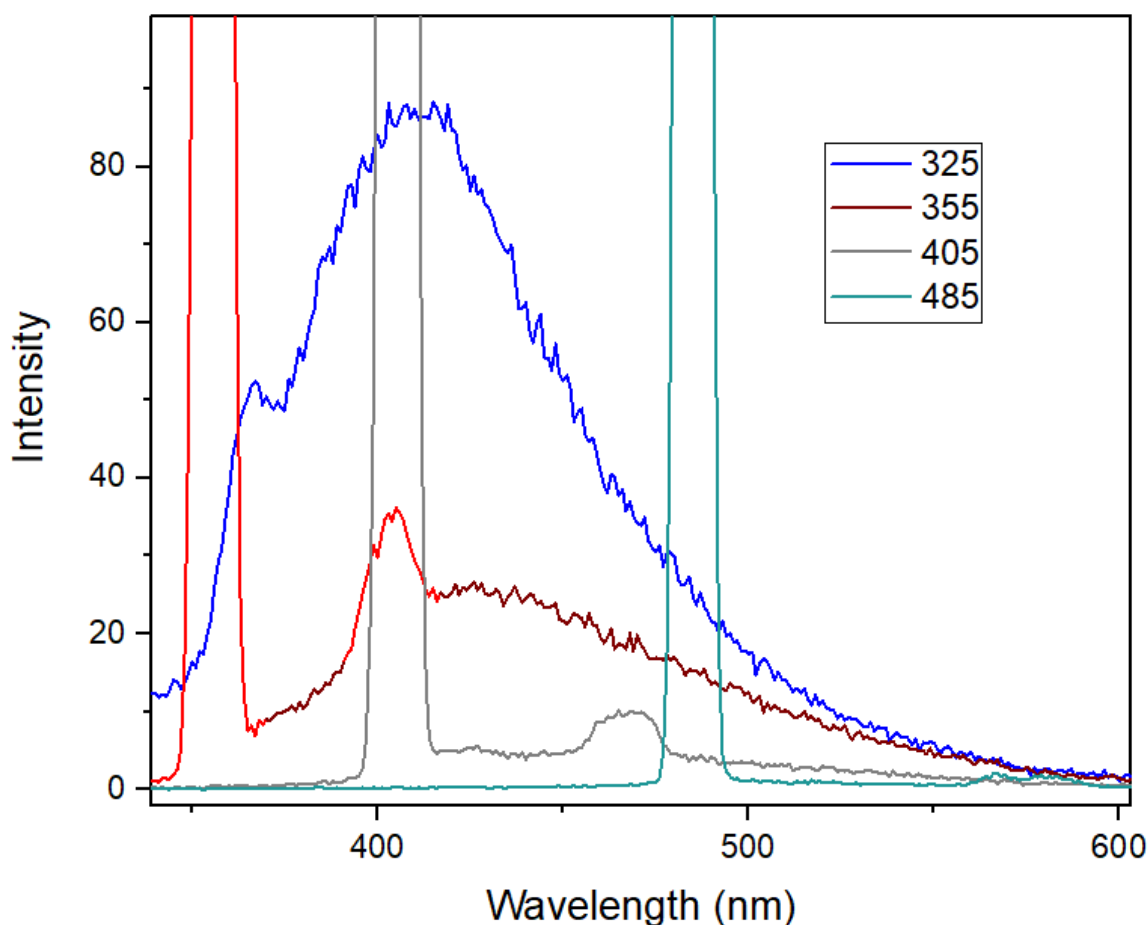


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146 *Figure S 3 TGA curves of uncalcined (black) and calcined SSZ-13 (red) showing the percentage of the original sample*
147 *weight as a function of the temperature at the sample*

148 Figure S3 shows TGA data for uncalcined and calcined SSZ-13 samples. Overall, the uncalcined sample
149 loses 21% of its total mass and the calcined sample loses 10%. The uncalcined sample is known to
150 contain organic templating agent so the higher overall mass loss is attributed to this. Below 250 °C
151 uncalcined sample does not experience significant mass loss, indicating that both that there is little
152 chemisorbed water due to the OSDA occupying acid sites and that the template is either broadly stable
153 or yet to decompose sufficiently to evacuate the pores up to this temperature. A modest mass loss is
154 visible at 150 °C during the isothermal portion of the temperature programme. The relative stability of
155 the trace up till 300 °C may be attributed to the initial fragmentation of the template, leading to the
156 formation of fragments that continue to be retained in the framework. Between 250 and 500 °C the
157 largest proportion of the total mass loss is seen, and this is attributed to the removal of organics. At
158 temperatures exceeding 500 °C and during the isothermal hold for 3 hours at 580 °C, additional mass
159 loss is seen, implying that in template material is progressively depleted at these higher temperatures.
160 A true detemplation would see the as-synthesised material being held at 550 °C for 10-12 hours,
161 ensuring as much organic material is removed as practically possible.

162 Unlike the uncalcined SSZ-13, the calcined sample experiences the majority of its mass loss below
163 250 °C. The rapid mass loss below 200 °C is attributed to water loss. Beyond this temperature, the
164 sample mass is exceedingly stable. Additionally, at 580 °C there is not any significant change in the
165 trace. This indicates that the sample was appropriately calcined as there is no major indication of
166 organic-related mass loss, and it indicates that the framework is thermally stable up to at least 580 °C.



168
169 *Figure S 4 Cary Agilent Spectrofluorometer data showing emissions spectra for trimethyladamantylammonium hydroxide at*
170 *325, 355, 405, and 485 nm excitation*

171 Spectrofluorometer data shows the fluorescence emission response of the templating agent
172 trimethyladamantylammonium hydroxide at selected wavelengths. At near-UV wavelengths like 325
173 nm, a clear fluorescence peak is seen at around 400 nm. With increasingly lower energy wavelengths
174 the fluorescence emission is shown to decrease with the response at imaging wavelengths of 405 and
175 485 nm (shown for 488 nm) being very low. It is expected that at 405 nm and 488 nm, therefore, that
176 the OSDA will not be directly visible.

177 **2.5 Lifetime Values**

178 *Table S1. Lifetime and amplitude values chabazite calcined at 550 °C for overall decay and four selected spatial regions of*
 179 *interest. A_i is the amplitude of the exponential component with exponential decay τ_i . $A_i\tau_i\%$ is the fractional amplitude,*
 180 *calculated as $A_i\tau_i/(A_1\tau_1 + A_2\tau_2 + A_3\tau_3)$, used to evaluate the weight of an exponential component. τ_{wavg} denotes the weighted*
 181 *average of the lifetimes τ .*

	ROI	A ₁	A ₂	A ₃	τ_1	τ_2	τ_3	A ₁ $\tau_1\%$	A ₂ $\tau_2\%$	A ₃ $\tau_3\%$	τ_{wavg}
Upper Plane	1	0.22	0.00	0.78	10.27	6.85	3.42	0.46	0.00	0.54	6.57
	2	0.21	0.48	0.31	12.59	4.98	0.99	0.50	0.45	0.06	8.53
Inner Plane	2	0.22	0.00	0.78	10.92	7.28	3.64	0.46	0.00	0.54	6.97
	3	0.25	0.00	0.75	16.62	11.09	5.46	0.50	0.00	0.49	11.09
	4	0.21	0.07	0.73	15.08	10.05	5.02	0.42	0.09	0.49	9.69
	5	0.07	0.29	0.64	10.01	2.98	0.58	0.36	0.45	0.19	5.07

182 *Table S2. Table of lifetimes and amplitudes for chabazite calcined at 630 °C in Figure S3 for top plane and central plane*
 183 *both overall and in each region of interest. A_i is the amplitude of the exponential component with exponential decay τ_i . $A_i\tau_i\%$*
 184 *is the fractional amplitude, calculated as $A_i\tau_i/(A_1\tau_1 + A_2\tau_2 + A_3\tau_3)$, used to evaluate the weight of an exponential component.*
 185 *τ_{wavg} denotes the weighted average of the lifetimes τ .*

	ROI	A ₁	A ₂	A ₃	τ_1	τ_2	τ_3	A ₁ $\tau_1\%$	A ₂ $\tau_2\%$	A ₃ $\tau_3\%$	τ_{wavg}
Upper Plane	1	0.95	0.00	0.05	0.40	3.37	2.98	0.71	0.00	0.29	1.14
	2	0.15	0.18	0.67	18.25	12.17	6.08	0.30	0.25	0.46	11.19
Inner Plane	3	0.20	0.05	0.75	11.40	7.80	2.66	0.49	0.08	0.43	7.35
	4	0.14	0.02	0.84	8.91	6.37	1.22	0.51	0.06	0.43	5.45

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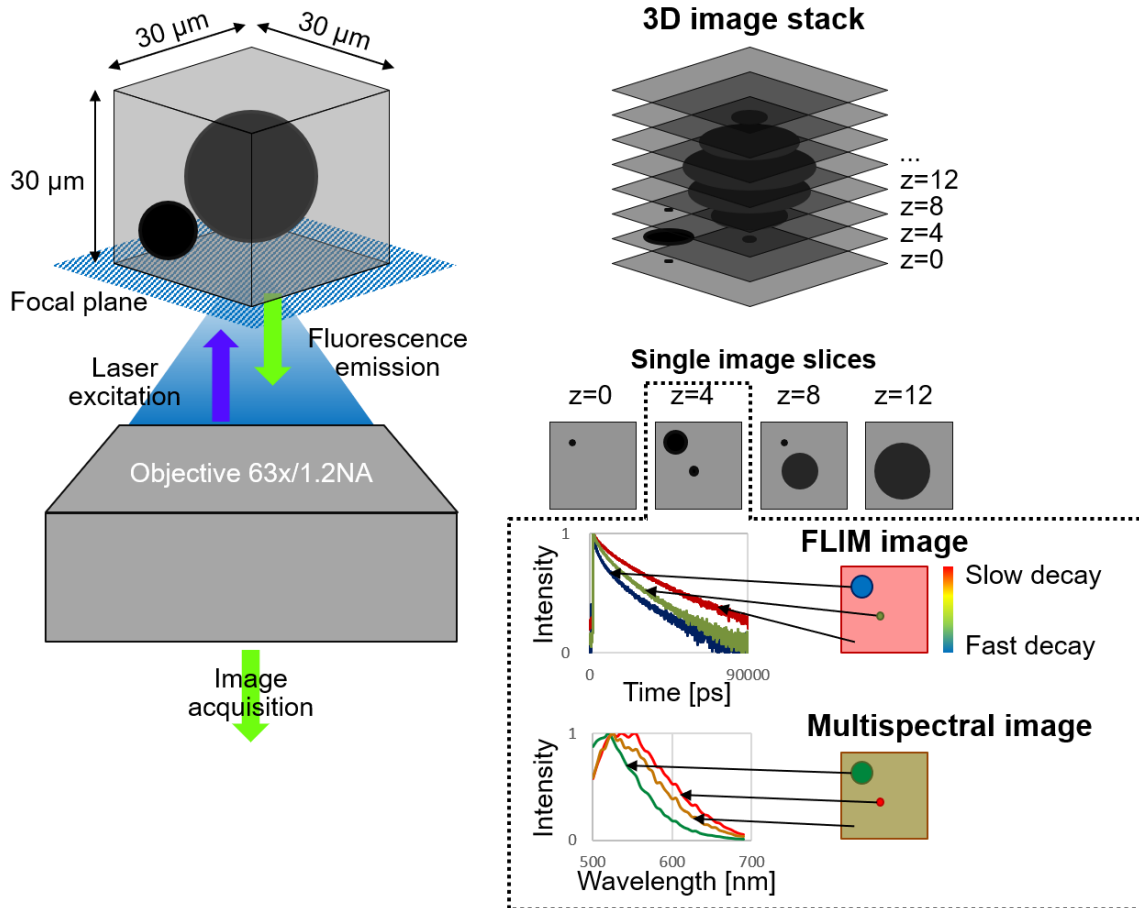
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189 2.6 Additional Confocal & FLIM Images

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191 Figure S5 shows a schematic diagram of how z-stack images were acquired, as well as the

192 methodology for selecting regions of interest for spectral analysis.



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194 *Figure S 5 . Scheme of principle of the Leica SP8 3D confocal acquisition. In this microscopy technique, a sample is*
195 *illuminated with a point scanning laser while the fluorescence is collected by an objective lens and sent to a detector to form*
196 *an intensity-based image. A pinhole is placed in front of the detector to reject out-of-plane fluorescence and scattered light.*
197 *By moving the sample through the focal plane, one can acquire a stack of images at varying sample depths in order to later*
198 *reconstruct a 3D image. The same principle is used to acquire FLIM and multispectral images. In these cases, every pixel*
199 *contains either decay information (FLIM image) or spectral information (multispectral image).*

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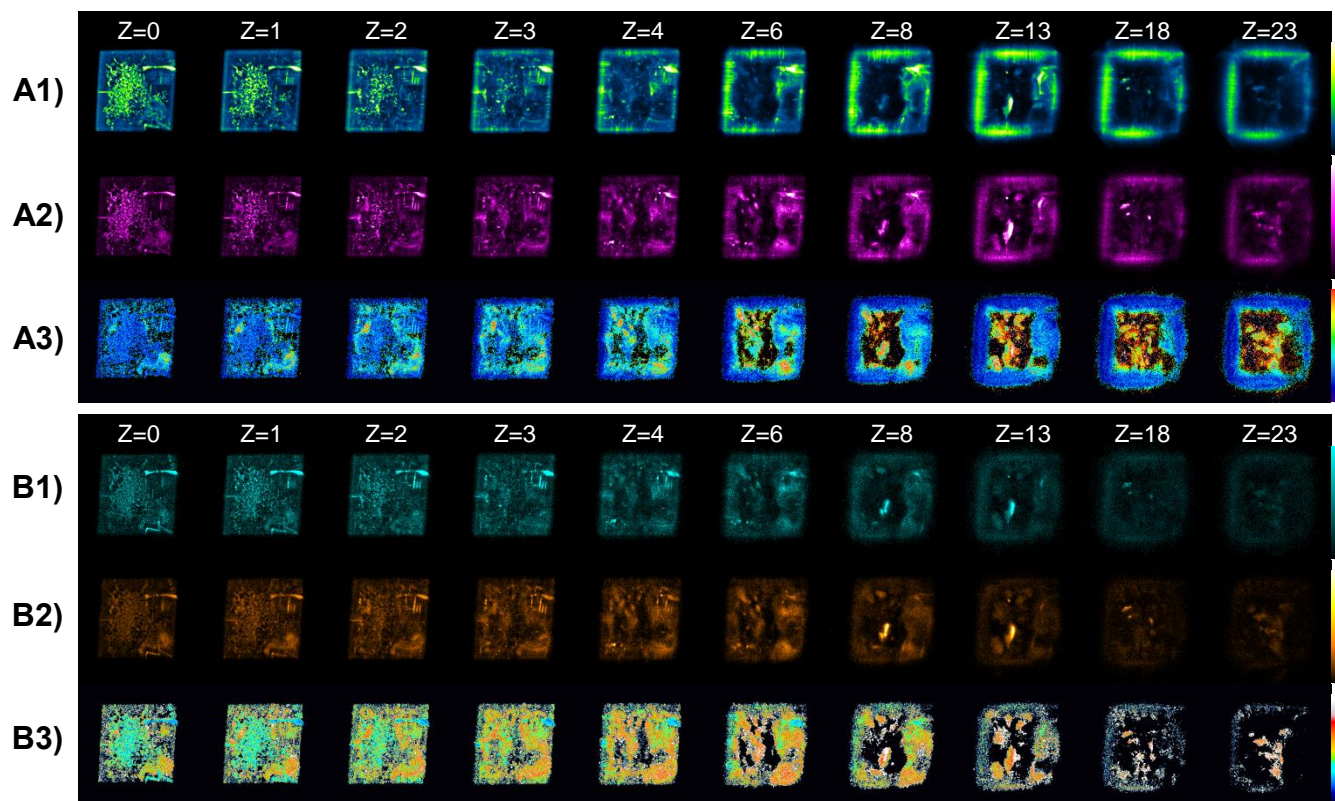
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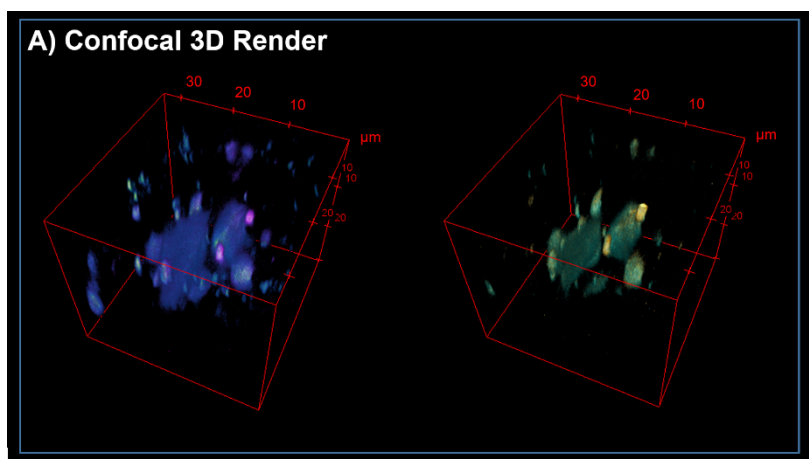
212 Additional confocal z-stack slices and 3D constructions are presented here. Figure S6 shows further z-
213 stack planes for two-colour confocal imaging. The ratio images have been calculate as channel in the
214 longer wavelength range divided by the channel in the shorter wavelength range; the ratio shows in
215 which areas the red emission is more pronounced compared to the shorter wavelength one.

216



217 *Figure S6 . Selected z-stack planes for two-colour confocal images of chabazite calcined at 550 °C. A) Single planes acquired*
218 *under 405 nm laser excitation at different depths in the 450 to 550 nm emission channel (1), in the 580 to 700 nm emission*
219 *channel (2) and the ratio image between the second and the first channel (3). B) Single planes acquired under 488 nm laser*
220 *excitation at different depths in the 500 to 550 nm emission channel (1), in the 580 to 700 nm emission channel (2) and the*
221 *ratio image between the second and the first channel (3).*

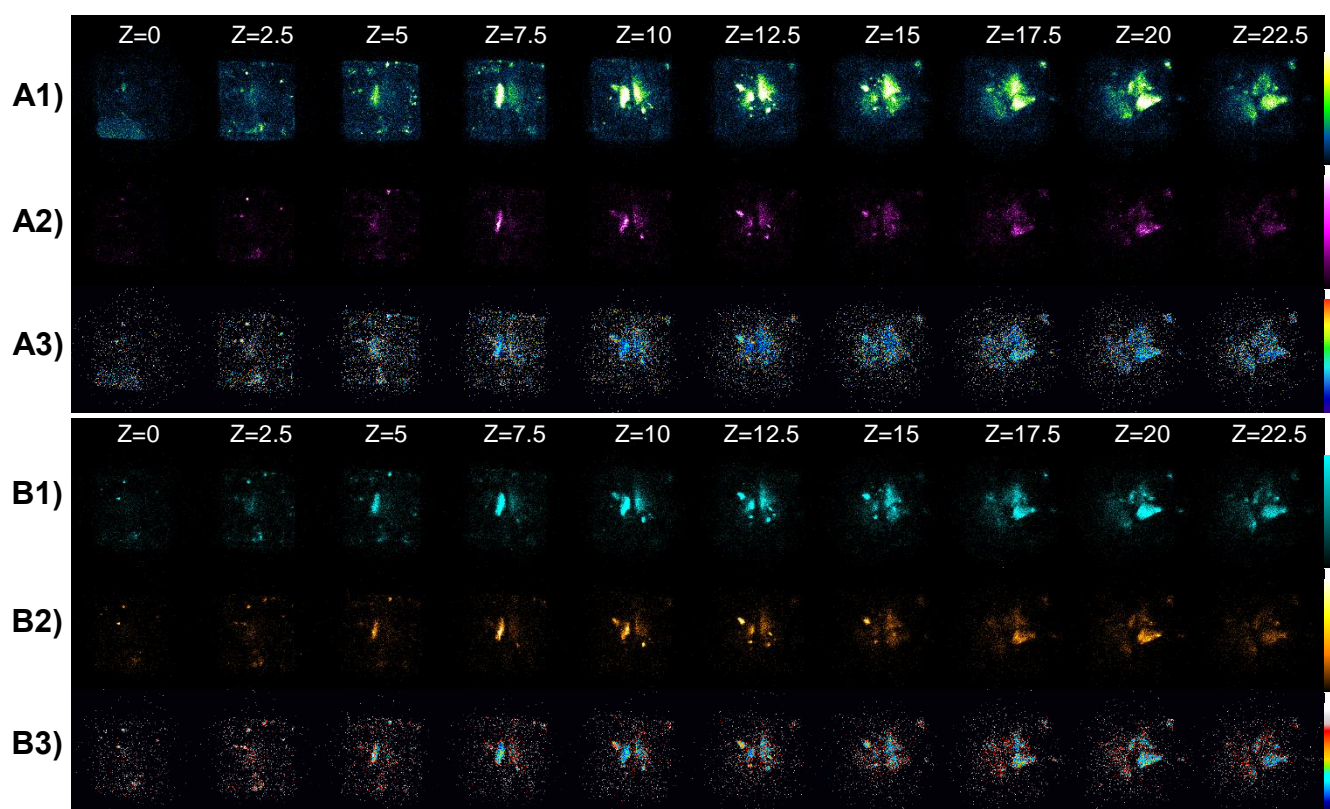
222 Figure S7 shows 3D reconstructions of confocal and FLIM z-stacks for chabazite calcined at 630 °C.
223 Compared to chabazite calcined at 550 °C, the 630 °C has less emission across the crystal faces. As a
224 result, in the 3D renders the internal occlusions are more visible. A concentration of emissive regions
225 is seen localised clearly in the centre of the crystal. Figure S8 shows selected planes from the z-stack.



226

227 *Figure S 7 3D reconstructions for chabazite calcined at 630 °C obtained from a 2-colour stack of 120 slices acquired every*
 228 *0.35 μm at A) $\lambda_{\text{ex}}=405 \text{ nm}$ where the blue-green channel captures emission from 450-550 nm and is merged with the*
 229 *magenta channel capturing emission from 580-700 nm; B) $\lambda_{\text{ex}}=488 \text{ nm}$ where the cyan channel captures emission from 500-*
 230 *550 nm and is merged with the orange channel capturing emission from 580-700 nm.*

231



232 *Figure S 8 Selected z-stack planes for two-colour confocal images of chabazite calcined at 630 °C. A) Single planes acquired*
 233 *under 405 nm laser excitation at different depths in the 450 to 550 nm emission channel (1), in the 580 to 700 nm emission*
 234 *channel (2) and the ratio image between the second and the first channel (3). B) Single planes acquired under 488 nm laser*
 235 *excitation at different depths in the 500 to 550 nm emission channel (1), in the 580 to 700 nm emission channel (2) and the*
 236 *ratio image between the second and the first channel (3).*

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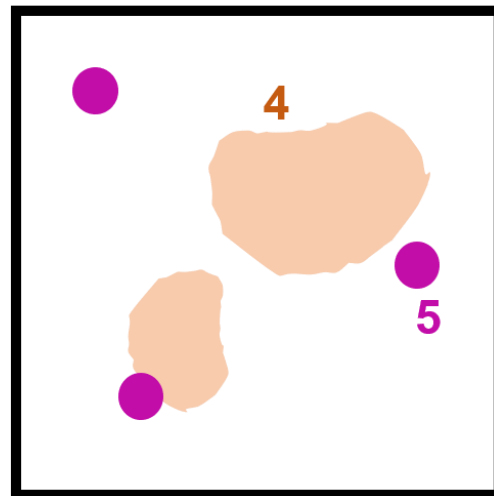
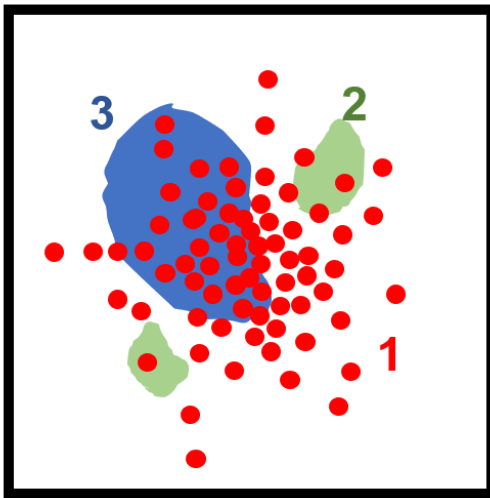
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241 **3 Schematic Diagram of Deposits**

242 Figure S9 is a schematic representation of the types of emissive deposits seen in the two samples. Both
243 samples show evidence of two types of subsurface deposits, one of which is smaller with significant
244 emission intensity (2 or 5) and another which appears as larger regions (3 or 5). Seen only in the 550 °C
245 sample are the distribution of smaller agglomerates (1) across the crystal faces that appear to be confined
246 to the first few microns of the external surfaces. In 3D slices of inner planes these deposits can be seen
247 as fringing around the edge of the crystal perimeter.

A) Calcined at 550 °C

B) Calcined at 630 °C



248

249 *Figure S 9 Schematic diagram of different types of emissive deposits discriminated on the basis of their shape and lifetime.*

250

251 **4 References**

- 252 [1] I. Lezcano-Gonzalez, U. Deka, B. Arstad, A. Van Yperen-De Deyne, K. Hemelsoet, M.
253 Waroquier, V. Van Speybroeck, B. M. Weckhuysen, A. M. Beale, *Phys. Chem. Chem. Phys.*
254 **2014**, *16*, 1639
- 255 [2] J. Schindelin, I. Arganda-Carreras, E. Frise, V. Kaynig, M. Longair, T. Pietzsch, S. Preibisch,
256 C. Rueden, S. Saalfeld, B. Schmid, J.-Y. Tinevez, D. J. White, V. Hartenstein, K. Eliceiri, P.
257 Tomancak, A. Cardona, *Nat. Methods* **2012**, *9*, 676–682.
- 258 [3] S. C. Warren, A. Margineanu, D. Alibhai, D. J. Kelly, C. Talbot, Y. Alexandrov, I. Munro, M.
259 Katan, C. Dunsby, P. M. W. French, *PLoS One* **2013**, *8*, DOI 10.1371/journal.pone.0070687.
- 260 [4] M. A. Digman, V. R. Caiolfa, M. Zamai, E. Gratton, *Biophys. J.* **2008**, *94*, 14–16.
- 261 [5] J. Ofner, K. Kamilli, E. Eitenberger, G. Friedbacher, B. Lendl, A. Held, H. Lohninger, *Anal.*
262 *Chem.* **2015**, 9413.
- 263 [6] S. Mosca, A. Artesani, D. Gulotta, A. Nevin, S. Goidanich, G. Valentini, D. Comelli,
264 *Microchem. J.* **2018**, *139*, 500–505.
- 265 [7] H. Lohninger, J. Ofner, *Spectrosc. Eur.* **2014**, *26*, 6–10.
- 266 [8] C. J. Fahrni, *Curr. Opin. Chem. Biol.* **2007**, *11*, 121–127.
- 267 [9] S. Vogt, J. Maser, C. Jacobsen, *J. Phys. IV Fr.* **2003**, *104*, 617–622.
- 268 [10] Y. P. Hong, S. Chen, C. Jacobsen, *Proc. SPIE--the Int. Soc. Opt. Eng.* **2015**, 9592, 95920W.
- 269 [11] C. Baerlocher, L. McCusker, “Database of Zeolite Structures,” can be found under
270 <http://www.iza-structure.org/databases/>, **n.d.**

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273 Please email Andrew.beale@ucl.ac.uk for copies of the raw data which includes: Raman, FLIM and
274 XRF spectra.

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